

Radiological Health Data

VOLUME IV, NUMBER 5
May 1963

U.S. DEPARTMENT OF HEALTH, EDUCATION, AND WELFARE
Public Health Service

In August 1959, the President directed the Secretary of Health, Education, and Welfare to intensify Departmental activities in the field of radiological health. The Department was assigned responsibility within the Executive Branch for the collation, analysis and interpretation of data on environmental radiation levels. The Department delegated this responsibility to the Division of Radiological Health, Public Health Service.

Radiological Health Data is published by the Public Health Service on a monthly basis. Data are provided to the Division of Radiological Health by other Federal agencies, State health departments, and foreign governments. Pertinent original data and interpretive papers are invited from investigators. Accepted material will be appropriately credited. The reports are reviewed by a Board of Editiorial Advisors with representatives from the following Federal agencies:

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RADIOLOGICAL HEALTH DATA

VOLUME IV, NUMBER 5 MAY 1963

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U.S. DEPARTMENT OF HEALTH, EDUCATION, AND WELFARE
Public Health Service

• Division of Radiological Health



SECTION I.—AIR AND FALLOUT

Fission Product Beta Activity in Airborne Particulates and Precipitation

Quick and sensitive detection of fission product activity fluctuation in the environment is possible through a program of continuous surveillance of gross beta activity in air and precipitation. The information obtained through such surveillance does not by itself permit evaluation of biological effects due to fallout, but it does form the basis of an alerting system and can be used as a rough guide to when and where more extensive monitoring of radioactivity in food, milk, and water is desirable.

January 1963 gross beta concentrations are presented here in reports from the Radiation Surveillance Network, Canadian Radioactive Fallout Study Program, and the Mexican Radioactive Fallout Program. Because of differences in equipment and techniques, the results from one network are not directly comparable with those of another. However, some intercalibration factors have been determined in a study conducted by Lockhart and Patterson of the U.S. Naval Research Laboratory (1, 2). An application of the results of this study is presented in figure 3, in which the January gross beta in air data from Canada and the U.S. were combined in one contour map. To adjust to a common baseline, the U.S. data were multiplied by a factor of 1.54, the U. S.-Canadian intercalibration factor suggested by the NRL study.

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Lockhart, L. B. Jr., and R. L. Patterson, Jr.: Intercalibration of Some Systems Employed in Monitoring Fission Products in the Atmosphere, NRL Report 5850, Washington, D.C. (November 13, 1962).

(2) Lockhart, L. B. Jr., and R. L. Patterson, Jr.: Intercalibration of Some Air Monitoring Systems, Radiological Health Data, 3:466-70, Superintendent of Documents, Government Printing Office, Washington 25, D.C. (December 1962).

RADIATION SURVEILLANCE NETWORK January 1963

Division of Radiological Health, Public Health Service

The Radiation Surveillance Network (RSN) comprises 72 sampling stations distributed among the fifty States, Guam, and Puerto Rico (see figure 1). These stations are manned predominantly by State health department personnel.

Air

Daily 24-hour air samples are collected by a high-volume air sampler with a 4-inch diameter carbon-loaded cellulose dust filter. Field estimates of the gross beta activity of airborne particulates are derived by comparing portable survey meter readings of these filters with readings taken from a Sr⁹⁰–⁹⁰Y known activity source. This determination is usually made about 5 hours after collection to eliminate interference from naturally occurring radon daughters. The Network's station operators contribute to a daily national report by telephoning their field estimates to the Radiation Surveillance Center, Division of Radiological Health, Washington, D.C.

The filters are then forwarded to the Radiation Surveillance Network laboratory in Rockville, Maryland, for a more refined measurement using a thin-window gas-flow proportional counter. Each filter is counted at least 3 days after the end of the sampling period and re-counted 7 days later. The initial 3-day aging of the sample eliminates interference from naturally-occurring radon and thoron daughters. The two counts, separated by a 7-day



interval, are used to estimate the age of fission products and to extrapolate the activity to the time of collection. The extrapolation is performed by using the Way-Wigner formula $(AT^{1,2} = constant)$.* The daily concentrations and estimated age are reported by the PHS in a monthly RSN report (1).

The average fission-product beta concentrations in surface air during January 1963, as determined by laboratory analysis and extrapolated to the time of collection, are given in table 1.

In order to compare these data with the gross beta reported by Canada, an adjustment is required. The relationship given by Lockhart and Patterson¹ is

 $\frac{PHS}{Canada}$ = 0.65 \pm 0.048 (one standard deviation)

which may be written:

 $1.54 \times \text{PHS} \pm 7.4 \text{ percent} = \text{Canada}$

Because of the higher filter efficiency and lower self-absorption in the Canadian system compared with RSN, it was considered more appropriate to adjust the RSN values upward to correspond with Canada rather than to adjust the Canadian data downward.

The Canadian air data and the adjusted RSN data rendered the concentration contour map in figure 3, which includes most of the North American continent.

Precipitation

Continuous sampling for total precipitation is conducted at most stations on a daily basis using funnels with collection areas of 0.4 m². A 500-ml aliquot of the collected precipitation is evaporated to dryness, and the residue is forwarded to the laboratory to be counted by the same method used for analyzing the air samples, including extrapolation to the time of collection. If the collected sample is between 200 and 500 ml, the entire sample is evaporated. When a sample is smaller than 200 ml (equivalent to 0.5 mm or 0.02 inches of rainfall) the volume of precipitation is reported, but no analysis is made. January 1963 averages of gross beta activity in precipitation, expressed in micromicrocuries per liter (µµc/liter) and millimicrocuries per square meter (m_{\(\mu\cepc\)}c/m²), are presented in table 2.

See references (1) and (2) on page 225

^{*} In this expression, A is the activity and T is the time after fission product formation.

TABLE 1.—GROSS BETA ACTIVITY OF PARTICU-LATES IN AIR, RSN, JANUARY 1963

TABLE 2.—GROSS BETA ACTIVITY IN PRECIPITA-TION, RSN, JANUARY 1963

[Concentrations in µµc/m³]

8	Station location	Number of samples	Maxi- mum	Mini- mum	Averages
Alaska:	Adak Anchorage Attu. Fairbanks Juneau Kodiak Nome Point Barrow Island St. Paul Island		9.5 14 81 11 16 26 6.7 7.9 9.1	<0.10 <0.10 <0.10 0.53 <0.10 <0.10 <0.10 1.3 0.45	3.3 3.5 4.6 4.2 3.6 6.4 2.6 4.7
Ariz: Ark: Calif: Colo:	Phoenix Little Rock Berkeley Los Angeles Denver Hartford	29 28 30 21 29	25 15 22 23 21	5.3 1.7 0.53 7.5 6.8	13 9.3 10 15 10
Conn: Del: D.C: Fla: Ga:	Dover	30 22 31 29 30 31	11 19 14 20 20	0.56 3.5 1.2 4.1 1.6 0.19	8.8 5.7 11 8.8 5.8
Guam: Hawaii: Idaho: Ill: Ind: Iowa: Kans:	Agana Honolulu Boise Springfield Indianapolis Iowa City Topeka	30 30 30 29 28 28	11 20 9.4 11 8.4	1.7 3.1 1.4 0.57 0.99 1.7	5.1 9.0 5.6 6.0 4.8
Ky: La: Maine: Md:	Frankfort	29 29 31 19 22	24 16 12 8.2 12	2.7 0.59 1.8 1.3 0.45 3.0	11 7.6 6.8 4.4 5.4
Mass: Mich: Minn: Miss:	Lawrence	29 28 31 29 29 21	12 20 12 9.1	1.5 4.3 2.5 1.6 1.2 2.0	6.4 10 7.9 6.2 9.2
Mo: Mont: Nebr: Nev: N.H: N.J:	Jefferson City	30 30 19 28 20 31	12 15 13 35 17	1.2 3.6 1.3 5.6 4.7 0.31	6.2 8.8 5.9 17 9.6
N. Mex: N.Y: N.C: N. Dak:	Santa Fe	27 28 29 15 30 30	16 12 11 7.7 27	5.3 0.72 1.3 2.1 1.3 3.3	8.6 5.8 7.4 5.8 9.8 7.9
Ohio: Okla: Ore:	Cincinnati Columbus Painesville Oklahoma City Ponca City Portland	21 29 30 30 30 27	11 12 16 12 6.2	1.3 1.1 1.4 2.7 1.4 1.2	5.7 7.6 7.6 7.6 3.8
Pa: P.R: R.I: S.C: S. Dak: Tenn:	Harriaburg San Juan Providence Columbia Pierre Nashville	21 22 30 27 29 28	10 5.6 13 18 9.7 16	0.32 1.1 0.91 0.82 2.8 2.1	5.8 3.1 6.6 6.6
Tex: Utah: Vt: Va: Wash:	Austin El Paso Salt Lake City Barre Richmond Seattle	31 31 31 31 30 30	16 17 18 15 11 16	1.9 2.8 3.8 2.0 2.1 0.53	8.7 8.0 10 8.2 5.4 7.1
W. Va: Wisc: Wyo:	Charleston	22 31 27	13 13 20	2.1 2.2 2.9	7.2 7.0 7.1

^{*}Weighted average obtained by summing the products of individual sampling times and the corresponding activities, and dividing by the summation of the individual sampling times.
b Dash indicates no sample received.

	Station location	Average concen- tration (μμc/liter)	Total deposition (mac/m³)
Alaska:	Adak	*	
	Anchorage Attu	1,700	32
	Fairbanks	2,300	100
	Juneau	1,500	98
	Kodiak	*	*
	Nome Point Barrow Island		
	St. Paul Island	*	*
Ariz:	Dhamis		
Ark:	Phoenix Little Rock	3,500	64
Calif:	Berkeley	840	79
Colo:	Los Angeles	2,200	8.
Conn:	Denver	2,400	120
		-,	
Del: D.C:	Dover	2 000	96
Fla:	Jacksonville	2,000 1,400	94
-	Miami	3,200	27
Ga:	Atlanta	*	*
Guam:	Agana		
Hawaii:	Honolulu		*
daho:	Boine	2,700 3,300	9.
III:	Indianapolis.	3,300	7.
lowa:	Iowa City	2,000	45 22
Kans:	Topeka		*
Ky:	Frankfort	4 900	89
La:	New Orleans	4,300 1,300	180
Maine:	Augusta	1,300	120
Md:	Presque Isle Baltimore		*
Mu.	Rockville	3,100	* 2.
Mass:	Lawrence	2,400	120
Mich:	Winchester Lansing	1,900	140
Minn:	Minneapolis	2,600	12
Miss:	Jackson	1,900	78
	Pascagoula	-	
Mo:	Jefferson City	2,000	14
Mont: Nebr:	Helena Lincoln	2,600	22
Nev:	Las Vegas	8,000	* 28
N.H: N.J:	Concord		
N.J:	Trenton	3,400	13
N. Mex:	Santa Fe	5,600	57
V.Y:	Albany	520	9.
	Buffalo	510	10
V.C:	New York	670	56
N. Dak:	Bismarck	3,600	15
Ohio:	Cincinnati		
MIO.	Columbus	2.200	77
	Paincaville	2,200 9,700	220
Okla:	Oklahoma City	6,200	40
re:	Ponca City Portland	1,100 3,600	25 76
Pa: P.R:	Harrisburg San Juan	440	120
R.I:	Providence	1,700	130
S.C:	Columbia	970	150
S. Dak: Fenn:	Pierre Nashville	7,400	12
Γex:	Austin	3,400	36
Utah:	El Paso	3,400 11,000 5,900	24
Vtan:	Salt Lake CityBarre	2,200	100 170
Va:	Richmond	1,700	66
Wash:	Seattle	2,700	63
W. Va:	Charleston	2,600	48
Wisc:	Madison	760	17
Wyo:			

^{*} Indicates no evaporated sample received.

REFERENCE

(1) Radiation Surveillance Network: Monthly Tabulation of Findings, Division of Radiological Health, Public Health Service, Washington 25, D.C. (Distribution by official request).

CANADIAN RADIOACTIVE FALLOUT STUDY PROGRAM

January 1963

Department of National Health and Welfare, Ottawa, Canada

As part of its Radioactive Fallout Study Program (RFSP), the Radiation Protection Division of the Canadian Department of National Health and Welfare monitors air and precipitation at 24 stations located at airports (see figure 2). The sampling equipment at these stations is operated by meteorologists of the Meteorological Services Branch of the Department of Transport. Detailed discussions of the sampling procedures, methods of analysis, and interpretation of results of the radioactive fallout program are contained in the Department's reports (1-5).



FIGURE 2.—CANADIAN AIR AND PRECIPITATION SAMPLING STATIONS

Air

In the collection of air samples, about 650 cubic meters of air are drawn through a high-efficiency 4-inch-diameter filter during a 24-hour period. These filters are sent daily to the Radiation Protection Division Laboratory in Ottawa. At the laboratory, a 2-inch-diameter disk is removed from each filter and counted with a thin-end-window Geiger flow counter system, calibrated with a Sr⁹⁰-Y⁹⁰ standard. Four successive measure-

ments are made on each filter to allow for the presence of natural activities and for the decay of short-lived fission products. The result is extrapolated to the end of the sampling period. Canadian air data for January 1963 are given in table 3, and presented in conjunction with U. S. adjusted air data in the concentration contour map (figure 3).

Precipitation

The amount of radioactive fallout being deposited on the ground is determined from measurements on material collected in special polythenelined rainfall pots. After transfer of the water to the sampling container, the polythene liner is removed, packed with the sample, and sent to the laboratory. January 1963 precipitation data for Canada, including some radiochemical analyses, are shown in table 4.

TABLE 3.—FISSION PRODUCT GROSS BETA ACTIVITY IN AIR, RFSP, JANUARY 1963

[Average concentrations in µµc/m³]

Station	Number of samples	Maximum	Minimum	Average
Calgary	29	42.0	2.1	17.7
Coral Harbour	30	13.0	0.0	4.4
Edmonton	23	31.0	3.1	11.8
Ft. Churchill	30	8.2	1.0	4.8
Ft. William	30	18.4	2.2	9.4
Fredericton	81	19.8	0.8	9.6
Goose Bay	30	10.7	1.1	5.6
Inuvik	31	29.0	3.6	11.8
Montreal		25.0	4.9	11.5
Moosonee		29.5	2.3	9.0
Ottawa		27.0	4.2	12.4
Regina	30	31.0	3.9	13.6
Resolute	29	11.0	0.8	5.7
Saskatoon	31	36.0	4.2	13.9
Sault Ste. Marie		28.2	2.8	11.5
Shearwater	31	33.0	0.6	11.1
Torbay	31	26.2	1.0	10.1
Toronto	30	41.0	4.8	14.6
Vancouver	31	38.0	0.9	12.1
Whitehorse	30	26.0	2.6	9.6
Windsor	31	22.0	5.6	14.1
Winnipeg	30	20.0	2.7	10.0
Yellowknife	31	16.2	4.5	9.6
Average				10.6

Recent coverage in Radiological Health Data:

Period	Issue
Third quarter 1961	May 1962
Fourth quarter 1961	September 1962
First quarter 1962	October 1962
Second and third quarters 1962	January 1963
October 1962	February 1963
November 1962	March 1963
December 1962	April 1963

TABLE 4.—FISSION PRODUCT GROSS BETA ACTIVITY IN SURFACE AIR, RFSP, JANUARY 1963

Station	Total beta activity		Deposition of specific radionuclides for selected sample $(m_\mu e/m^2)$				
	μμc/liter	mµe/m²	Srss	Srm	Zr**	Ca137	Ba140
Calgary Coral Harbour Edmonton Fredericton	3,308 8,462 2,523 991	84.0 64.5 98.5 110.6	5.00	0.25	13.4	0.35	3.06
Fort Churchill Fort William Googe Bay Inuvik	1,287 4,478 613 3,139	31.1 54.6 67.3 114.0					
Montreal	1,594 1,111 1,674 757	66.4 83.0 63.7 49.1	6.35	0.25	9.86	0.47	2.50
Regina Resolute Saskatoon Sault Ste. Marie	6,545 4,618 2,877 1,234	29.9 140.7 39.4 65.5					
Shearwater Torbay Toronto*	1,415 1,652	204.5 252.5	11.9	0.51	44.5	1.12	7.20
Vancouver	2,181	87.5	10.4	0.47	11.2	0.73	3.3
Whitehorse Windsor Winnipeg Yellowknife	3,871 2,589 6,366 3,118	77.7 73.6 42.0 63.3	4.48	0.19	5.70	0.34	2.8
Average	2,887	83.0					

¹ All values corrected for decay back to end of collection month.

² Values for strontium-90, cesium-137, and zirconium-95 do not include the activities of their daughter isotopes, yttrium-90, barium-137, and niobium-95.

³ No sample.

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- (1) Bird, P. M., A. H. Booth, and P. G. Mar: Annual Report for 1959 on the Radioactive Fallout Study Program, CNHW (RPD-3), (May 1960).
- (2) Bird, P. M., A. H. Booth, and P. G. Mar: Annual Report for 1960 on the Radioactive Fallout Study Program, CNHW (RPD-4) (December 1961).
- (3) Mar, P. G.: Outline of Procedure for the Radiochemical

Analysis of Dried Milk Powders for Strontium and Yttrium, CNHW (RPD-5), (June 1, 1960).

- (4) Beale, J. and J. Gordon: The Operation of the Radiation Protection Division Air Monitoring Program, CNHW (RPD-11), (July 1962).
- (5) Booth, A. H.: The Calculation of Maximum Permissible Levels of Fallout in Air and Water and Their Use in Assess-ing the Significance of 1961 Levels in Canada, CNHW (RPD-21), (August 1962).

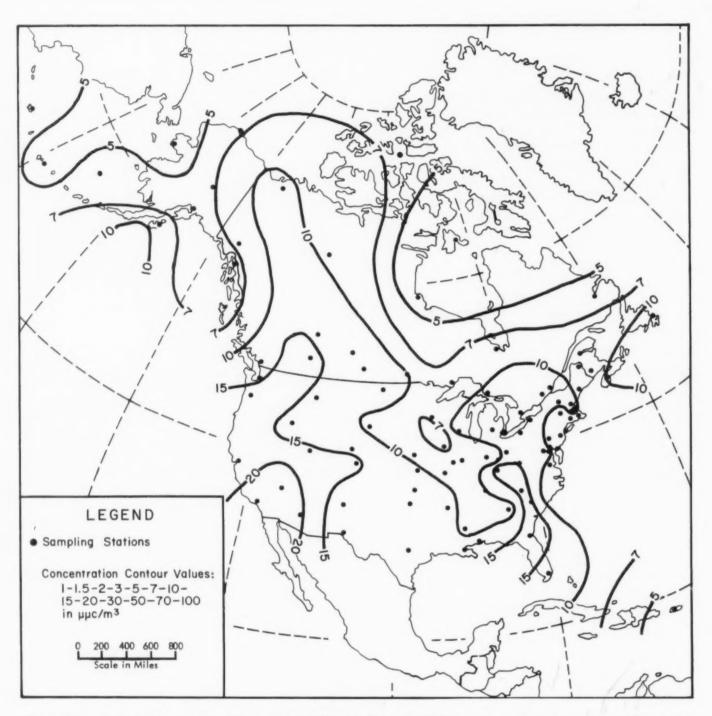


Figure 3.—AIRBORNE GROSS BETA CONCENTRATION CONTOURS FOR CANADA AND THE U. S. JANUARY 1963

MEXICAN RADIOACTIVE FALLOUT PROGRAM

January 1963

Radiological Protection Program National Commission of Nuclear Energy, Mexico

The Radiation Surveillance Network of Mexico was established by the Comision Nacional de Energía Nuclear (CNEN) through its Radiological Protection Program (RPP) in 1961 to provide a means for determining increased levels of radioactivity in air and precipitation due to fallout from nuclear tests.

Prior to the establishment of the network, two pilot sampling stations were set up in Mexico City and San Luis Potosí to aid in the selection of equipment and sampling sites. Since April 1962 the network has been expanded to twelve stations, eleven of which were in operation by the end of January 1963.

Eight of the twelve stations are located at airports and operated by airline personnel. The remaining four stations are located at Mexico City, Mérida, Veracruz, and San Luis Potosí. Staff members of the RPP operate the station at Mexico City, while the other three stations are manned by members of the Centro de Previsión del Golfo de México, the Chemistry Department of the University of Mérida, and the Instituto de Zonas Desérticas of the University of San Luis Potosí, respectively.

In December 1962, the station located in Tijuana was moved to Ensenada, a city on the Pacific Coast 60 miles from Tijuana, where it is operated by the staff members of the Escuela Superior de Ciencias Marinas of the University of Baja, California.

Sampling

The sampling procedure involves drawing air

24 hours a day, 3 or 4 days a week at the rate of approximately 1,200 cubic meters per day, through a 6" x 8" high-efficiency glass fiber filter using high-volume samplers. After each 24-hour period, the filter is removed and airmailed to the Laboratorio de Desechos Radiactivos (CNEN) in Mexico City for assay of gross beta activity. A minimum of 3 or 4 days after collection is allowed for decay of radon and thoron daughters. Data are not extrapolated to time of collection.

The maximum, minimum, and average fission-product beta concentrations in surface air during January 1963 are presented in table 5.



FIGURE 4.—FALLOUT NETWORK SAMPLING STATIONS IN MEXICO

TABLE 5.—GROSS BETA ACTIVITY OF AIRBORNE PARTICULATES, MEXICO, JANUARY 1963

[Concentrations in µµc/m³]

Station	Number of samples	Maximum	Minimum	Average
Acapulco	5	5.5	0.6	3.96
Ciudad Juárez	6	29.9	9.2	22.58
Ensenada	13	39.9	10.7	16.78
Guadalajara	12	14.2	1.8	5.78
La Paz	14	19.7	4.0	9.37
Méridà	15	42.6	3.3	12.9
México	12	4.2	2.1	3.3
San Luis Potosi	12	9.3	1.4 3.5	5.50
Torreón	12	38.5		18.96
Veracruz	21	17.8	0.4	4.1
Tuxtla Gutierres	19	15.4	0.15	4 5



SECTION II.—MILK

Milk Surveillance

Produced and consumed on a regular basis, milk is convenient to handle, is easily analyzed, and representative samples of milk consumed in any given area are easily obtained. Although milk is only one of the many sources of dietary intake of radionuclides, it is the single food item most often used as an indicator of the population's intake of radionuclides from the environment because fresh milk is consumed by a large segment of the U. S. population and contains most of the radionuclides identified as being biologically important.

PASTEURIZED MILK NETWORK January 1963 and Annual Summary for 1962

Division of Radiological Health and Division of Environmental Engineering and Food Protection, Public Health Service

JANUARY 1963

Raw milk studies were started by the Public Health Service in 1957 to determine relationships between dairy practices and radionuclide levels in milk. However, it became evident that the milk actually consumed by the population should be included in a broader sampling program. In 1960 the pasteurized milk network was initiated to provide data representative of the milk consumed in selected municipalities. Both raw and pasteurized milk sampling and analytical data were reported concurrently until June 1961 to permit comparison of the differences between the earlier.

limited, milkshed sampling results and those exhibited in pasteurized milk monitoring. Since June 1961, raw milk sampling has been carried out for investigative rather than monitoring purposes.

During January 1963, pasteurized milk surveillance was conducted at 62 Pasteurized Milk Network stations with the cooperation of State and local milk sanitation agencies. These samples, preserved with formaldehyde, are sent to the PHS Southwestern, Southeastern, and Northeastern Radiological Health Laboratories for analyses. Data from gamma analyses (iodine-131) are available to State public health officials and the Federal Radiation Council approximately 3 to 6 days after sample collection for possible public health action. Publication in Radiological Health Data follows 3 to 4 months after sample collection because of the time required for shipment, processing, radioanalysis of strontium, data compilation, and publication procedures.

Sampling and Compositing Procedures

The current program emphasizes (1) measurement of the concentrations of radioactivity in samples of pasteurized milk consumed by the public in various regions of the country, and (2) provision of at least one sampling point within virtually all States; additional points to be established when the requirements are indicated by widely varying conditions of the milk supply or when needed in order to cover large population groups. Each sample is a composite of subsamples which are specified to be collected from each plant

in proportion to the relative volumes of milk sold. Each of the 62 stations reflects from 80 to 100 percent of the milk consumed in that city, with the network average being about 90 percent. Prior to September 15, 1961, the composite of a sample was taken from one day's sales per month and was as representative of a community's total supply as could be achieved under practical conditions. Since the resumption of nuclear weapons testing, the sampling frequency has been increased. During January 1963, most stations were sampled twice a week. All surveillance data are subject to continuing review and evaluation to observe unusual patterns or concentrations which may require immediate attention. Further atmospheric nuclear testing may require re-evaluation and adjustment of the sampling frequency and analytical schedule for this program.

Minimum Reportable Concentrations

Iodine-131, cesium-137, and barium-140 are determined by gamma scintillation spectroscopy, while strontium-89 and strontium-90 are determined by radiochemical procedures. Minimum reportable concentrations used, in units of $\mu\mu c/liter$ are: Sr⁸⁹, 5; Sr⁹⁰, 1; I¹³¹, 10; Cs¹³⁷, 5; and Ba¹⁴⁰, 10.

Milk Collection and Distribution Times

A number of factors influence the concentration of radionuclides in milk. One factor (which may not be the most important) is the decay of radionuclides, particularly the short-lived ones, due to transfer times involved in collection and distribution of milk.

¹ See page 139 of March Radiological Health Data.

Table 1 gives a rough estimate of the times involved in the movement of pasteurized whole milk through the various stages from production to consumption. The headings in table 1 may be explained by the following statements. "Milking to processing plant" is the time required for storage on the farm and pickup and delivery to the processing plant. "In Processing plant" includes the time for storing raw milk in the processing plant, pasteurizing, transferring the milk through the plant. and storing the processed milk in cartons. "Processing plant to consumer purchase" is the time for delivering the milk from the processing plant to retail outlets and storing in the retail store. "Consumer purchase to consumption" is the time the milk is kept by the consumer before consumption in the usual liquid form. Thus, "Total time, milking to consumption" is the estimated total time between the time of milking and the consumption of the pasteurized fluid milk.

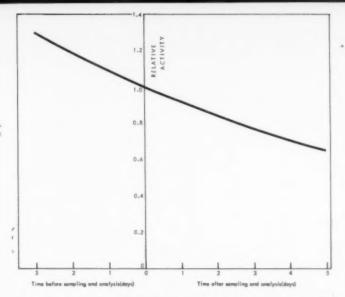
The ranges shown in table 1 do not imply that some milk processing plants distribute milk that is fresher than others. The chain of milk production involves assembling small quantities of milk from many individual farms into larger volumes for processing. The assembled milk is a composite of the times shown in the range. Consequently, part of the milk in a carton will be fresher than another part. Modern standards of biological cleanliness are high and the actual age difference between the fractions of milk in the carton has little bacteriological significance.

However, the total elapsed time is important in assessing the radiation dose from a short-lived nuclide such as iodine-131 ($T_{\frac{1}{2}}=8.05$ days). Figure 1 presents graphically the effect of the radioactive decay of iodine-131. Although analyses of milk samples are performed 1 to 4 days after collection, the data presented in table 2 have been

TABLE 1.—NORMAL TIMES REQUIRED FOR THE MOVEMENT OF PASTEURIZED WHOLE MILK FROM THE FARM TO THE CONSUMER 1

		Time in hours required for:										
	Milking to processing plant	In processing plant	Processing plant to consumer purchase	Consumer purchase to consumption	Total time, milking to consumption							
Range	2-44	1-8	0-48	0-72	3-148							
Average	23	4.5	24	36	75.5							

¹ Data supplied by the Milk Industry Foundation.



 $\begin{array}{ll} \textbf{FIGURE 1.} - \textbf{RELATIVE DECAY OF IODINE-131} \ (\textbf{HALF-LIFE} = 8.05 \ \textbf{DAYS}) \end{array}$

extrapolated to the time of collection. Samples are usually collected from cartons at or near the end of the processing period.

Data Presentation

Table 2 presents summaries of all available analyses for January 1963 (December 30, 1962-

January 26, 1963). When a radionuclide is reported by a laboratory as being below the minimum detectable concentration, one-half of this value is used in calculating the monthly average. A similar procedure is used for the network average. Although no data are presented on the stable potassium concentrations in milk, analysis has indicated that the usual range of concentrations is from 1.4 to 1.7 grams/liter. In January, for example, 12, 18, 24, and 4 stations reported respective monthly average potassium concentrations to be 1.4, 1.5, 1.6, and 1.7 grams/liter.

Figures 2-4 are concentration contour maps showing the estimated radionuclide concentrations in milk over the entire country. The value printed beside each station is the monthly average concentration for that station.

Discussion of Data

In January 1963 the amount of iodine-131 appearing in milk (figure 2) over the entire country changed abruptly from that for the previous month. During December 1962 most of the country had monthly average concentrations from 20 to

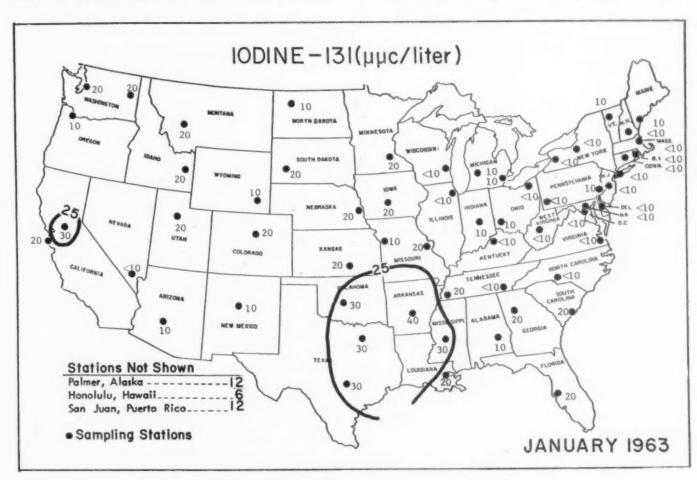


FIGURE 2.—IODINE-131 CONCENTRATIONS IN PASTEURIZED MILK

TABLE 2. RADIOACTIVITY IN PASTEURIZED MILK, JANUARY 1963

[Average radioactivity concentrations in µµc/liter]

		Calcium	(g/liter)	Stront	ium-89	Stront	ium-90	Iodin	e-131	Cesiun	m-137	Bariu	m-140
	Sampling locations	Fourth quarter	Avg. for month	Fourth quarter	Avg. for month	Fourth quarter	Avg. for month	Fourth quarter	Avg. for month	Fourth quarter	Avg. for month	Fourth quarter	Avg. for month
Ala: Alaska: Ariz: Ark: Calif:	Montgomery Palmer Phoenix Little Rock Sacramento San Francisco	1.26 1.33 1.32 1.26 1.32 1.34	1.28 1.24 1.26 1.23 1.29 1.34	40 80 25 120 20 20	80 25 25 125 15 60	15 13 5 29 4 4	15 12 4 29 8 5	40 110 20 110 20 20 20	10 20 10 40 30 20	35 60 20 70 20 20	45 85 20 70 25 25	20 30 <10 30 10 <10	10 10 <16 20 <10
Colo: Conn: Del: D. C: Fla:	Denver	1.34 1.13 1.11 1.20 1.25 1.24	1.34 1.12 1.16 1.21 1.24 1.25	20 40 50 40 30 55	10 5 10 10 30 85	11 11 15 18 10 17	13 12 16 13 12 18	20 40 80 60 50 60	20 <10 <10 <10 20 20	55 65 70 55 125 55	70 70 75 55 100 70	<10 10 20 20 10 20	<10 <10 <10 <10 <10 <10
fawaii: daho: ll: nd: owa: Kans:	Honolulu	1.26 1.31 1.12 1.17 1.36 1.34	1.19 1.25 1.12 1.16 1.28 1.32	20 25 45 50 75 55	40 10 <5 10 20 30	6 13 12 13 14 13	6 13 14 15 14 12	20 40 70 80 110 70	20 20 <10 10 20 20	40 75 65 55 50 40	45 75 75 65 60 50	<10 10 10 10 20 20	<1 1 <1 <1 <1 <1 <1
Ky: La: Maine: Md: Mass: Mich:	Louisville New Orleans Portland Baltimore Boston Detroit Grand Rapids	1.23 1.28 1.15 1.22 1.13 1.13	1.23 1.27 1.09 1.22 1.16 1.16 1.18	95 70 50 40 60 50 35	30 160 <5 <5 5 5 10	23 24 17 17 19 15	21 30 20 15 18 16 15	80 70 50 60 50 70 50	<10 20 10 <10 <10 10	45 60 110 60 100 75 65	55 75 105 60 100 75 75	40 20 20 20 20 20 20 10	<10 2 <1 <1 <1 <1 <1 <1
Minn: Miss: Mo: Mont: Nebr:	Minneapolis Jackson Kansas City St. Louis Helena Omaha	1.35	1.24 1.34 1.28 1.28 1.30 1.36	70 90 100 55 45	15 170 30 20 25 25	23 19 17 14 16 15	20 25 16 12 14	70 70 120 60 80 90	20 30 10 20 20 20	95 45 45 45 45 85	105 55 45 60 90 60	30 30 30 20 30 20	<1 2 <1 1 1 2 1
Nev: N. H: N. J: N. Mex: N. Y:	Las Vegas	1.16 1.12 1.33 1.10 1.13	1.19 1.16 1.14 1.28 1.11 1.09	25 50 45 20 40 55 45	10 5 <5 20 10 <5 <5	6 18 13 5 13 17	6 20 14 6 15 16	10 40 60 30 40 80 50	<10 <10 <10 10 <10 <10 <10 <10	40 115 65 25 80 75 65	40 110 65 30 90 70	<10 10 20 20 10 20 10	<1 <1 <1 <1 <1 <1 <1 <1
N. C: N. D: Ohio: Okla:	Charlotte Minot Cincinnati Cleveland Oklahoma City Portland	1.34 1.16 1.16 1.21	1.25 1.27 1.13 1.18 1.24 1.32	50 60 60 50 65 150	25 15 20 5 65 70	19 27 17 14 17 18	25 28 18 15 17 12	20 60 90 70 100 60	<10 10 10 <10 30 10	50 65 45 60 40 80	50 90 55 60 50 70	20 20 10 <10 20 40	<1 <1 <1 <1
Pa: P. R: R. I: S. C: S. D:	Philadelphia Pittsburgh San Juan Providence Charleston Rapid City	1.13 1.14 1.17 1.11 1.26	1.12 1.15 1.24 1.15 1.24 1.14	45 55 75 45 50 50	10 <5 150 10 60 25	14 19 10 16 20 15	18 19 12 14 20 15	80 110 30 60 40 70	10 <10 30 <10 20 20	60 80 45 85 60 70	65	10 20 10 20 20 20 20	<1 <1 <1 <1
Tenn: Tex: Utah: Vt:	Chattanooga Memphis Austin Dallas Salt Lake City Burlington	1.25 1.23 1.23 1.35	1.24 1.28 1.31	95 85 35 90 30 55	70 95 70 130 15 5		22 22 9 18 12 16	60 80 90 150 40 60	20	50 40 30 45 75 85	45 25 55 95	30	<
Va: Wash: W. Va: Wis: Wyo:	Norfolk Seattle Spokane Charleston Milwaukee Laramie	1.23 1.36 1.40 1.22 1.18	1.25	45	10		18 9 13 19 9	50 60 30 40 90 20			75 90 45 65	10 20 20	< < <
	average	-		-	-	-	-	61		-	_	-	

 $200~\mu\mu c/liter$, while in January 1963 most of the stations reported from <10 to 20 $\mu\mu c/liter$. The Mississippi Valley area was the area with the greatest concentrations during both months. December 1962 iodine-131 concentrations of 100 to 330 $\mu\mu c/liter$ were observed in this area while January 1963 averages were from 20 to 40 $\mu\mu c/liter$.

The monthly average strontium-89 concentrations in milk for January 1963 are similar to those in December 1962 in most areas (see figure 3). In January 1963 strontium-89 concentrations generally were below $50~\mu\mu c/liter$.

Monthly average strontium-90 concentrations in milk for January 1963 were also similar to the ones in December 1962. Most stations reported

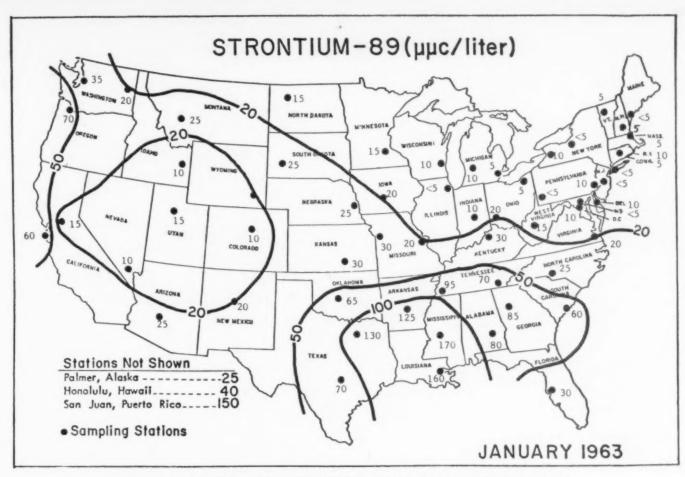


FIGURE 3.—STRONTIUM-89 CONCENTRATIONS IN PASTEURIZED MILK

concentrations from 10 to 20 $\mu\mu$ c/liter (see figure 4). In the Southwest, strontium-90 concentrations below 10 $\mu\mu$ c/liter were observed, while the lower Mississippi Valley, together with a region extending eastward through the Carolinas to the coast, experienced concentrations of 20 to 30 $\mu\mu$ c/liter. North Dakota and Minnesota also experienced approximately this same range of concentrations.

As mentioned previously, radiostrontium levels did not change appreciably since the previous month, but iodine-131 concentrations in milk dropped sharply. Examination of barium-140 and cesium-137 concentrations for December 1962 and January 1963 shows that barium-140 concentra-

tions in milk decreased while cesium-137 concentrations remained approximately constant. In summary, the concentrations of the "long-lived" nuclides remained approximately constant while the concentrations of "short-lived", nuclides decreased from December 1962 to January 1963. Concentrations of strontium-89 (T $_{\rm i}=50.5~{\rm days})$ in milk showed a slight decrease.

These observations might be explained by one or more of the following presumptive factors:

- 1. A relatively low total fallout occurred in January 1963 as compared with December 1962.
- 2. The relative proportion of short to longlived radionuclides in January fallout was less than that in December fallout.

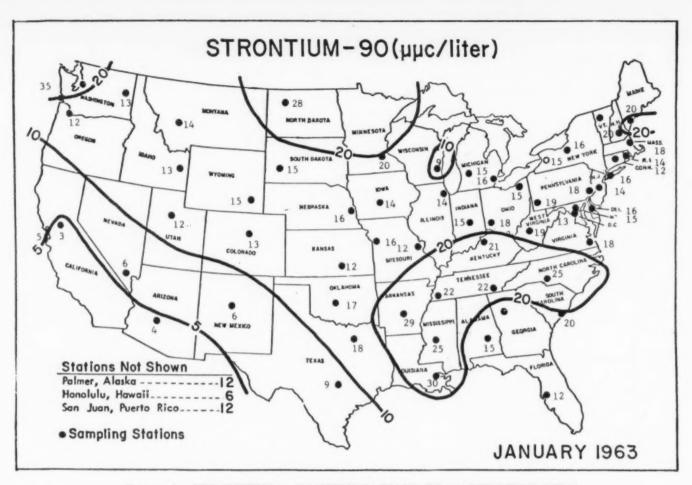


FIGURE 4.—STRONTIUM-90 CONCENTRATIONS IN PASTEURIZED MILK

3. In January, cows consumed feed which contained relatively little fresh fallout as opposed to December (e.g., relatively more stored feed than pasturage was used).

Deposition data were not available to confirm or disprove that factors 1 and 2 were operative. However, additional information secured concerning factor 3 indicates that, during the first week of January, mild weather permitted livestock throughout the Great Plains Region to feed from Pastures (1). In the lower Ohio Valley, the lower Mississippi Valley, and eastward there was limited grazing due to damage to winter pastures. During the latter three weeks of January, snow and

low temperature conditions forced cattle from the pasture and pushed supplemental feeding to high levels throughout the nation.

Selected Monthly Strontium-90 Profiles

Continuing the practice of previous issues of *RHD*, figure 5 presents the average monthly strontium-90 concentrations in pasteurized milk from 16 additional cities in the sampling program. Each individual graph shows the strontium-90 concentrations in milk from one city in each of the four U.S. Bureau of Census regions.

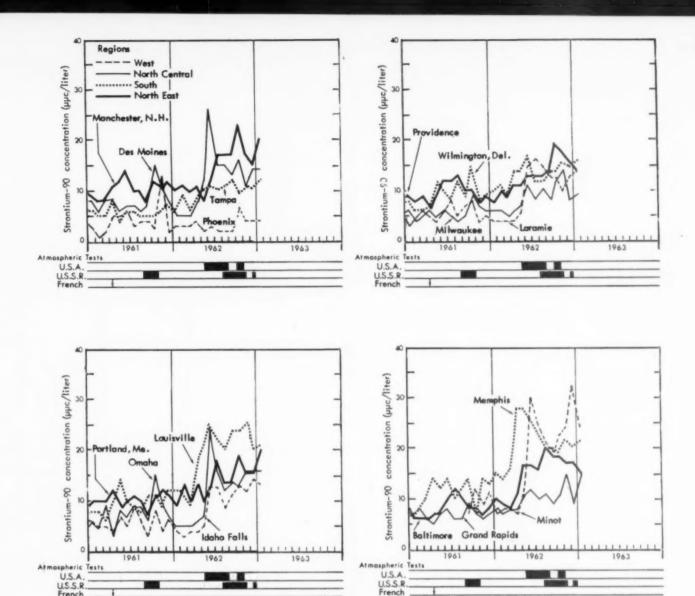


FIGURE 5.—STRONTIUM-90 CONCENTRATIONS IN PASTEURIZED MILK

ANNUAL SUMMARY FOR 1962

The annual average radionuclide concentrations in the Pasteurized Milk Network have been summarized for calendar year 1962 and appear in table 3 below. Included are values for stable calcium and five nuclides of primary importance. Comparable tabular data for 1960 and 1961 have been published previously (2).

The annual average values in the table have been compiled from weekly as opposed to monthly averages. This procedure is dictated since each "month", for which monthly radionuclide averages have been published in 1962 contains either four or five weeks depending on the number of full weeks ending within each calendar month. This avoids a possible bias introduced by averaging unequal units of time (months). The year 1962 extends from

January 1, 1962 through December 29, 1962. In averaging, when a radionuclide concentration is reported as being below the minimum detectable value, one half of this value is used in the calculations. Where an "average" is determined below this minimum detectable value, the minimum detectable value preceded by a "less than" sign is the reported average.

Figures 6-11 show the geographical distribution of strontium-89 and iodine-131 for 1961 and 1962, and strontium-90 for 1961 and 1962. The values shown on the figures at each station represent the yearly average for that station.

Discussion

The 1960 average for strontium-89 was below the detectable limit since there was no atmospheric nuclear testing since 1958. With the resumed test-

TABLE 3.—ANNUAL SUMMARIES OF RADIOACTIVITY IN PASTEURIZED MILK, 1962

[Average radioactivity concentrations in µµc/liter]

	Sampling locations	Calcium (g/liter)	Strontium-89	Strontium-90	Iodine-131	Casium-137	Barium-140
Ma:	Montgomery ^a	1.21	59	15	18	37	
laska:	Palmer	1.14	51	10	104	37	
riz:	Phoenix	1.13	18	8	12	12	<
rk:	Little Rock	1.21	133	29	37 12	62	<
alif:	Sacramento	1.15 1.17	19 32	5	14	17	
olo:	Denver	1.17	29	10	15	38	<
onn:	Hartford	1.14	22	11	21	45	<
el:	HartfordWilmington	1.14	36	13	32	49	<
. C:	Washington	1.17	35	14	23	40	
la: a:	Washington Tampa Atlanta	1.23 1.21	25 84	10 18	16 23	108 57	<
awaii:	Honolulu	1.11	27	5	12	27	<
daho:	Idaho Falls	1.14	24	9	25	44	
1:	Chicago	1.14	30	11	38	39	<
nd:	Indianapolis	1.17	38	13	83	35	<
owa:	Des Moines	1.15	65	12	59	32	
ans:	Des Moines	1.15	50	11	59	27	
y:	Louisville	1.19	76	19	29	33	
a: faine:	New Orleans	1.24	170 30	30	23 22	86	<
laine: [d:	Portland	1.18	33	14 15	24	45	<
fass:	Baltimore	1.15	31	16	22	76	
lich:	Detroit	1.14	27	11	35	44	<
icn.	Detroit	1.19	23	10	26	42	<
linn:	Minneapolis	1.14	55	16	43	58	
lien:	Jackson	1.27	154	24	23	56	
lo:	Jackson Kansas City	1.14	84	14	84	29	
	St. Louis	1.14	53	13	33	32	
font:	Helena		44 57	12	40 53	50 32	
lebr:	Omaha		51	13	50	02	
lev:	Las Vegasb			-	-	-	
I. H:	Manchester	1.15	28	14	20 22	87	
J. J:	Trenton	1.13	27 17	11	21	39 14	4
I. Mex:	Buffalo	1.15 1.12	24	11	24	50	
	New York	1.12	29	14	32	51	
	Syracuse	1.14	26	ii	27	40	
V. C:	Charlotte	1.23	56	19	9	39	
J. Dak:	Minot		45	18	41	49	
hio:	Cincinnati	1.16	48	14	39	30	
	Cleveland Oklahoma City	1.15	29	12	31	38	
kla:	Oklahoma City	1.17	70	17	48	37	
re:	Portland	1.18	82	13	27	53	
a:	Philadelphia		28	12	29	45	
	Pittsburgh		34	15	40	53	
P. R:	San Juan	1.16	71	10	14 23	40 58	
R. I:	Providence	1.14 1.22	25 68	12 19	17	56	
. C: . Dak:	Rapid City	1.10	54	15	39	46	
enn:	Chattanooga	1.25	117	22	22	56	
	Memphis	1.22	105	21	27	28	
'ex:	Austin	1.18	28	7	28	20	
v. v	Dallas	1.20	76	15	49	34	
tah:	Salt Lake CityBurlington	1.16	30 30	9	87 23	52 58	
7a:	Norfolk		47	17	18	49	
Vash:	Seattle	1.16	59	15	27	61	
	SeattleSpokane	1.19	36	12	59	49	
V. Va:	Charleston	1.18	57	18	19	35	
Vis:	Milwaukee	1.17	23	8	39	37	
Wyo:	Laramie		52	10	55	56	
	average	1.17	50	13.4	32	45	

Averages for calcium (g/liter) and for barium-140 are for 11 months. No data for February 1962.
 No averages are given for Las Vegas since the station has not been in the Pasteurized Milk Network for the entire year.
 Averages are for 11 months. No data for November 1962.

ing of nuclear devices in September 1961, strontium-89 reappeared, as shown in the 1961 average values. Most of the U.S. reported averages for strontium-89 of below 20 µµc/liter. The States immediately west of the Mississippi River however, averaged yearly concentrations of strontium-89 of 20 to 30 $\mu\mu$ c/liter. In 1962, the concentrations were higher. Most of the country had yearly strontium-89 concentrations of 20 to 101 µµc/liter.

The Southwest reported values below 20 µµc/liter, while in the lower Mississippi valley concentrations of 100 to 170 $\mu\mu$ c/liter were recorded.

The 1960 average strontium -90 concentrations were 13 $\mu\mu$ c/liter throughout the entire country. The Southwest showed a yearly average of below $5 \mu\mu c$ /liter. The resumption of atmospheric nuclear testing did not appreciably increase the yearly average strontium-90 content of milk for 1961.

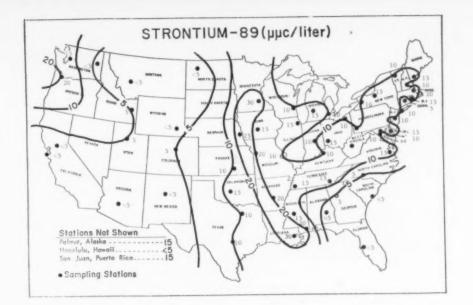
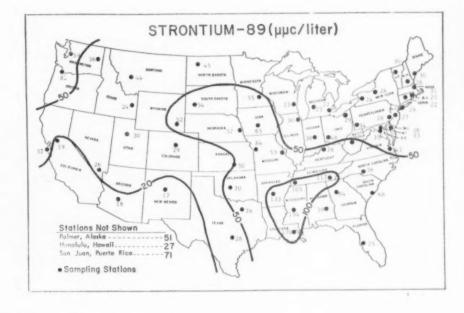


FIGURE 6.—ANNUAL AVERAGE STRONTIUM-89 CONCENTRA-TIONS IN PASTEURIZED MILK, 1961

FIGURE 7.—ANNUAL AVERAGE STRONTIUM-89 CONCENTRA-TIONS IN PASTEURIZED MILK, 1962



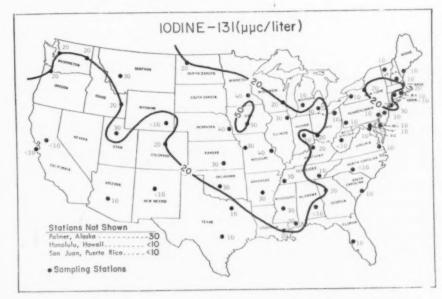


FIGURE 8.—ANNUAL AVERAGE IODINE-131 CONCENTRATIONS IN PASTEURIZED MILK, 1961

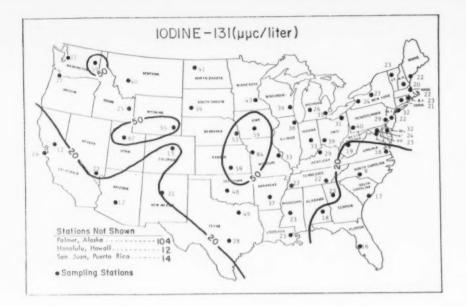
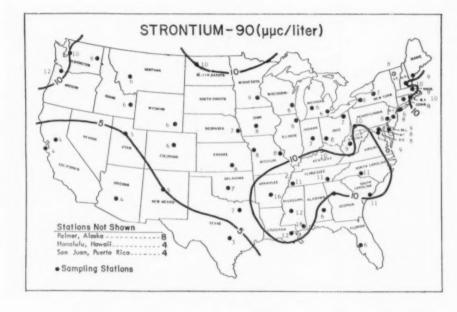


FIGURE 9.—ANNUAL AVERAGE IODINE-131 CONCENTRATIONS IN PASTEURIZED MILK, 1962

FIGURE 10.—ANNUAL AVERAGE STRONTIUM-90 CONCENTRA-TIONS IN PASTEURIZED MILK, 1961



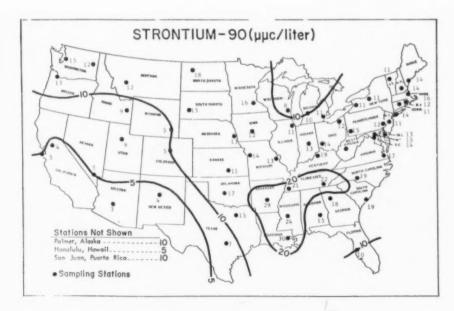


FIGURE 11.—ANNUAL AVERAGE STRONTIUM-90 CONCENTRA-TIONS IN PASTEURIZED MILK, 1962

The national pattern for 1961 appears similar to that for 1960 with the exception that for 1961 one station had an average concentration of 16 $\mu\mu c/$ liter; this was the highest annual average concentration observed that year. In 1962, strontium-90 levels generally increased. The lower Mississippi Valley region exhibited a yearly average concentration of 20 to 30 $\mu\mu c/$ liter. Values in the Southwest were below 5 $\mu\mu c/$ liter; and the balance of the nation reported values intermediate to these.

Iodine-131 was not detectable in milk in 1960. Iodine-131 appeared in the 1961 average with one station reporting a value of 50 $\mu\mu$ c/liter, the highest annual average for that year. In 1962, eight sta-

tions reported yearly average iodine-131 levels above 50 $\mu\mu$ c/liter, the highest being 104 $\mu\mu$ c/ liter. Iodine-131 yearly average concentrations differ markedly from the monthly values. While large excursions in the monthly average concentration of iodine-131 occur at various stations, very large monthly differences between stations and between regions were averaged out over the year.

REFERENCES

(1) United States Department of Agriculture Statistical Reporting Service: Weekly Weather and Crop Bulletin, Volume L, Nos. 1-4, Superintendent of Documents, Government Printing Office, Washington 25, D.C. (January 1963, price per year \$3.00, price per single copy 10 cents.

(2) Public Health Service; Milk Monitoring Program, Radiological Health Data, 3:186-91 (June 1962).

CALIFORNIA MILK NETWORK July-September 1962

State of California, Department of Public Health

Surveillance of the concentrations of specific radionuclides in milk is one phase of California's Department of Public Health program of radiation control. This milk monitoring function has been conducted since 1960 by the Department's Bureau of Radiological Health, a constituent of the Division of Environmental Sanitation.

The surveillance program involves the weekly sampling of milk from 11 major milksheds (see figure 12). Originally, radioanalyses were performed for potassium-40, cesium-137, and strontium-90 and calcium. More recently, interest has broadened to include not only the above but also iodine-131, strontium-89, and barium-lanthanum-140. All milk is now being analyzed for these three radionuclides only.

Results and Discussion

Table 4 presents strontium-89 and strontium-90 monthly averages for the 11 milksheds from July to September 1962. It should be noted that the units used in table 1 are $\mu\mu$ c/g calcium. Comparisons of other data in terms of $\mu\mu$ c/liter may be made using the approximate equivalent value of 1.1 grams of calcium per liter of milk. No iodine-131 data are given because all samples indicated that iodine-131 concentrations in milk



FIGURE 12.—CALIFORNIA MILKSHEDS

were below the detectable limit of 10 $\mu\mu$ c/liter during the third quarter of 1962.

Del Norte and Humboldt Counties have reported milk with higher strontium-89 and strontium-90 concentrations than the other areas. Strontium-90 concentrations in milk have remained relatively constant from July through September, but strontium-89 concentrations underwent a significant decrease from July through August in all milksheds.

TABLE 4.—RADIONUCLIDES IN CALIFORNIA MILK, JULY-SEPTEMBER 1962

[Concentrations in µµc/g calcium]

Sampling area		Strontium-89		Strontium-90				
	July	August	September	July	August	September		
Alameda	22.8	2.9	a	3.9	2.3			
Del Norte	61.4	44.4	50.3	16.8	19.7	13.8		
Fresno	9.8	3.1	2.8	3.4	1.8	1.9		
Humboldt	20.3	14.1	21.3	4.9	5.6	6.0		
Los Angeles	11.2	2.2	2.2	1.5	1.8	2.1		
Mendocino	17.8	9.9	4.3	2.9	3.8	3.6		
Sacramento	14.1	3.9	3.2	2.4	2.3	1.9		
San Diego	12.2	1.5	1.8	1.8	2.3	1.8		
Santa Clara	12.3	3.4	1.8	1.9	1.3	2.0		
Shasta	24.0	8.6	5.5	3.8	3.6	3.1		
Sonoma	13.1	6.0	3.9	2.8	3.1	2.3		

^a Dash indicates no sample.

CONNECTICUT MILK NETWORK April 1960–December 1962

Connecticut State Department of Health

The Connecticut State Department of Health began monitoring pasteurized fluid milk for strontium-89 and strontium-90 in April 1960. During this initial period a monthly composite milk sample, representative of the milk sold in the central area of the State, was collected from the processing plant and analyzed for radiostrontium. Beginning in May 1962, the Department's milk monitoring program was expanded to include the determination of iodine-131, barium-lanthanum-140, arel cesium-137 in milk samples collected weekly. Since September 1962, the frequency of this sampling has been twice a week.

In addition to sampling milk sold in the central area of the State during 1961, a three-month composite sample representative of the milk sold in the southeastern section of the State was also collected and analyzed for radiostrontium.

From May 1962 to November 1962, weekly samples were collected from milk dealers in five areas of the State (figure 13). These samples, together with the regular composite sample from the central section of the State, were analyzed for the

gamma-emitting nuclides. Collection and analysis of these dealer samples will be reinstated when the cows return to pasture feeding in the spring.

A low-background Geiger counter with a thin end-window and gas-flow chamber is used for counting radiostrontium following chemical separation. Initially, the iodine-131, barium-lanthanum-140, and cesium-137 were determined by gamma-scintillation spectrometry using a 4" x 4" sodium iodide crystal and a single-channel analyzer. In November 1962, a 400-channel analyzer was acquired.

TABLE 5.—YEARLY AVERAGES OF STRONTIUM-89 AND STRONTIUM-90 IN MILK FROM CENTRAL AND SOUTHEAST AREAS OF CONNECTICUT, APRIL 1960-DECEMBER 1962

[Concentrations in ##c/liter]

Year	Stront	ium-89	Stront	ium-90
	Central	Southeast	Central	Southeast
April-December 1960 1961	* 6.1 21.0	17.2	8.1 7.3 9.4	10.7

^{*} Dash indicates not detectable.

The 1960, 1961 and 1962 yearly averages of strontium-89 strontium-90 concentrations in milk are given in table 5. The monthly averages for the data obtained from May 1962 through December 1962 are reported in table 6.

TABLE 6.—RADIONUCLIDES IN CONNECTICUT MILK, MAY-DECEMBER 1962

[Concentrations in µµc/liter]

	Central area* composite samples			Central area dealer samples				Southwest area dealer samples		South central area dealer samples		Northeast area dealer samples								
Month	Srss	Sr **	I131	Ba- La ¹⁴⁰	Cs137	I131	Ba- La ¹⁴⁰	Ca137	I131	Ba- La ¹⁴⁰	Cs137	I131	Ba- La ¹⁴⁰	Cs137	Inn	Ba- La ¹⁴⁰	Ca137	I 131	Ba- La ¹⁴⁰	Csu
May	30 49 34 24 27 34 39	8.0 14.4 13.9 11.3 11.5 10.4 9.5 8.8	10 <10 <10 <10 110 90 50	<10 <10 20 10 20 30 <10 <10	80 80 90 70 60 60 70 70	<10 <10 10 150 80 60	10 <10 20 40 30 10	90 100 80 70 60 70	10 <10 50 70 120	<10 20 40 40 30	120 140 60 90 100	<10 <10 <10 10 80 60	10 10 10 10 20 10	90 80 90 60 70 60	<10	10	90	20 160 70 100	<10 30 40 10	11: 6: 10: 9:

^{*} Samples also analyzed by Public Health Service laboratory.

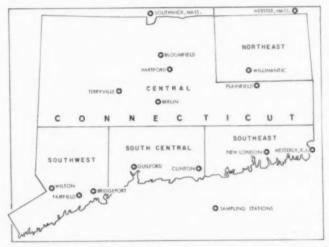


FIGURE 13.—CONNECTICUT MILK SAMPLING STATIONS

INDIANA MILK NETWORK January–February 1963

Bureau of Environmental Sanitation Indiana State Board of Health

The Indiana State Board of Health began sampling pasteurized milk for radiological analyses in September 1961. Indiana was geographically divided into five major milksheds, and one large dairy within each milkshed was selected as a sampling station (see figure 14).

The milk samples are routinely analyzed for iodine-131, cesium-137, barium-lanthanum-140, strontium-89, and strontium-90. Analyses for the gamma emitters iodine-131, cesium-137, and barium-lanthanum-140 are conducted on a weekly basis except when iodine-131 results exceed 100 $\mu\mu$ c/liter, at which time the frequency of sampling is increased. Strontium-89 and strontium-90 anal-

yses are performed monthly on samples which are composited from weekly aliquots.

An ion exchange analytical procedure (1) is used for strontium-89 and strontium-90 analyses. A 512-channel pulse height analyzer and shielded 4 x 4-inch sodium iodide crystal are used for the gamma analysis of iodine-131, cesium-137, and barium-lanthanum-140.

The monthly averages of the data obtained for the individual sampling stations and the State averages are reported in table 7. The State average is an arithmetic average of the station values.

REFERENCE

(1) Porter C., D. Cahill, R. Schneider, P. Robbins, W. Perry, and B. Kahn: Determination of Strontium-90 in Milk by an Ion Exchange Method, Analytical Chemistry 33:1306-8 (September 1961).



FIGURE 14.—INDIANA MILK SAMPLING LOCATIONS

TABLE 7.—RADIONUCLIDES IN INDIANA MILK, JANUARY-FEBRUARY 1963

[Concentrations in µµc/liter]

Sampling location	Strontium-89		Strontium-90		Iodine-131		Cesium-137		Barium- Lanthanum-140	
	January	February*	January	February*	January	February	January	February	January	February
Evansville Fort Wayne Indianapolis Rochester Seymour	20 b— <10 <10 <10		17 18 23 22 24		<10 <10 <10 <10 <10	<10 <10 <10 <10 <10	50 50 55 55 50	40 55 50 60	<10 <10 <10 <10 <10	<10 <10 <10 <10 <10
State average	10		21		<10	<10	50	55	<10	<10

NEW YORK MILK NETWORK November-December 1962

Division of Special Health Services State of New York Department of Health

Milk samples, collected routinely from six cities -Albany, Buffalo, Massena, Newburgh, New York City, and Syracuse (figure 15), are analyzed for radionuclide content by the State of New York Department of Health. The pasteurized milk samples are collected from processing plants daily and composited weekly for strontium-89, strontium-90, and iodine-131 analyses. The sampling practice at Albany differs in that milk is collected from a marketing point. Also, at Albany and New York City daily samples are analyzed for iodine-131.

In the event that a city reports iodine-131 concentrations exceeding 100 pc/liter, increased surveillance is undertaken. The matrix method (1) is used for the analysis of spectral data to determine the concentrations of gamma-emitting nuclides in milk. With this method, the individual nuclide contributions to the gamma spectrum are separated by solution of simultaneous equations describing the spectral interferences.

The analytical procedure for strontium-89 and strontium-90 is based on ion exchange methods to concentrate the strontium, elutriation of strontium isotopes from the ion exchange resin with sodium chloride, gathering by means of sodium carbonate, isolation by means of ethylenediaminetetraacetic acid (EDTA), and counting of radiostrontium with a low background beta counter having a 0.8 mg/

cm² window. The strontium-90 portion is differentially estimated by a second count 40 hours later to determine the rate of growth of its daughter product, yttrium-90.

Table 8 shows the monthly radionuclide concentration averages for November and December 1962. The numbers in parentheses below the data values are the number of determinations used to compute the monthly average.

REFERENCE

(1) Kahn, B., et al.: Rapid Methods for Estimating Fission Product Concentrations in Milk, Public Health Service Pub. No. 999-R-2. Single free copies may be obtained from Public Inquiries Branch, PHS, U.S. Department of Health, Education, and Welfare, Washington 25, D.C.



FIGURE 15.—NEW YORK MILK SAMPLING LOCATIONS

<sup>Available next month.
Dash indicates no sample.</sup>

TABLE 8.—RADIONUCLIDES IN NEW YORK MILK, NOVEMBER-DECEMBER 1962

[Concentrations in pc/liter]

Sampling	Strontiu	m-89	Strontiu	ım-90	Iodine-	131	Cesium-137	
location	Nov.	Dec.	Nov.	Dec.	Nov.	Dec.	Nov.	Dec.
Albany	30 a(4)	11 (4)	11 (4)	9 (4)	25 (25)	<20 (15)	65 (25)	62 (15
Buffalo	30	13 (2) 15	(4)	8	<20 (6)	<20 (4)	62 (6)	(15 83 (4
Massena	(4) 35 (4)	(4)	12 (4) 10	(4) 12 (4) 8	<20 (4) 43	<20 (4)	117 (4) 77	107
Newburgh	(4) 51 (4)	17 (4)	(4)	(4)	43 (3) 55	<20 (4) 25	(5)	52
New York City	46 (5)	17 (4)	(5)	10 (4)	(14)	(9)	b	_
Syracuse	(3)	12 (4)	7 (3)	6 (4)	<20 (3)	<20 (4)	44 (5)	64

a Numbers in parentheses indicate the number of samples upon which the monthly average is based.

b A dash indicates no sample or no analysis.

OREGON MILK NETWORK November 1962–February 1963

Division of Sanitation and Engineering Oregon State Board of Health

Oregon's milk surveillance network was organized in March 1962, by the Oregon State Board of Health in cooperation with the Oregon State Department of Agriculture and Oregon State University. The half-gallon samples of packaged homogenized pasteurized milk are collected on a monthly basis by the Oregon Department of Agriculture and the City of Portland from eight milk districts throughout the State (see figure 16). Sampling on a weekly basis is performed when iodine-131 concentrations in milk exceed 100 pc/ liter. The samples are forwarded to the Oregon State Board of Health radiation laboratory for iodine-131, cesium-137, and barium-140 analyses. The analyses are performed with a 3" x 3" sodium iodide scintillation crystal and a 512-channel pulseheight analyzer. All milk samples, after being gamma-scanned, are frozen and stored for future strontium-90 analyses.

Table 9 presents the Oregon milk surveillance data for November 1962 through February 1963. The Portland composite sample represents contributions from nearly all milk sheds in Oregon, including some in Washington State. Thus, it tends to represent a State average. Tillamook has



FIGURE 16.—OREGON PASTEURIZED MILK NET WORK SAMPLING LOCATIONS

shown the highest concentrations with a maximum iodine-131 concentration for a single sample of 530 pc/liter on November 12, 1962. The frequency of sampling at Tillamook was temporarily increased to a weekly rate when concentrations exceeded 100 pc/liter. A review of the values obtained at Tillamook established that the iodine concentrations decreased at a rate corresponding to an 8-day half-life. This indicated an apparent single intrusion of fission products, rather than a continuous contribution from the atmosphere. The Tillamook and Coos Bay samples tend to be higher than other locations, probably due to the greater

yearly precipitation along the Pacific Coast compared to the inland areas.

To maintain a check on the analytical procedures and instrument calibration, a split sample of the official U.S. Public Health Service pasteurized milk sample from Portland is obtained on a weekly basis for analysis. These data are compared to the Public Health Service data in figure 17.

Table 9.—RADIONUCLIDE CONCENTRATIONS IN OREGON MILK, NOVEMBER 1962-FEBRUARY 1963

[Average concentrations in pc/liter]

Sampling location	Number of samples analyzed			Iodine-131			Cesium-137			Barium-140						
	Nov.	Dec.	Jan.	Feb.	Nov.	Dec.	Jan.	Feb.	Nov.	Dec.	Jan.	Feb.	Nov.	Dec.	Jan.	Feb
Baker Coos Bay Eugene Medford Nyssa Ortland (composite) Ortland (local producer)	1 0 1 1 1 4 4	1 2 1 1 1 5 5	1 1 1 1 1 4 4	1 1 1 1 1 3	<15 50 40 20 110 40	30 60 15 <15 15 40	<15 30 <15 <15 <15 15	<15 30 <15 <15 <15 <15 <15	45 45 35 35 90 70	50 110 50 35 90 75 70	40 110 45 30 40 60 70	45 110 65 45 30 70 65	<15 <15 <15 <15 <10 20	<15 70 <15 <15 <15 <15	<15 <15 <15 <15 <15 <15	<1 <1 <1 <1 <1 <1 <1
tedmond	1 4	1 2	1	1	80 390	<15 100	15 30	<15 <15	40 245	30 155	35 115	60 80	<15 220	<15 <15 60	<15 <15 <15	1

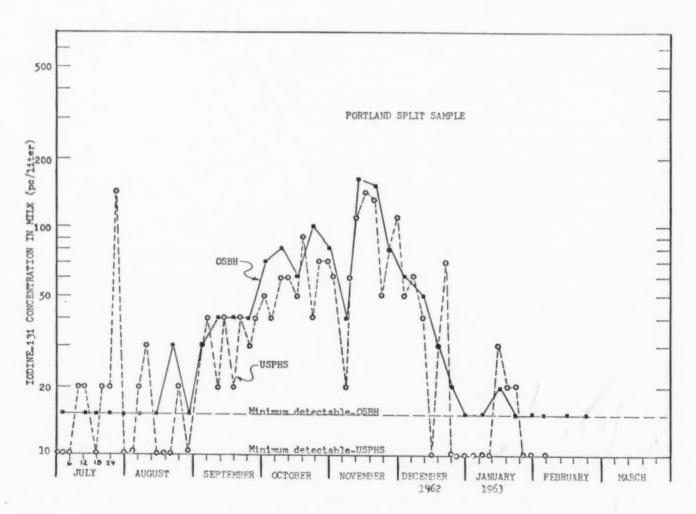


FIGURE 17.—COMPARISON OF PUBLIC HEALTH SERVICE AND OREGON STATE BOARD OF HEALTH MILK ANALYSES FOR PORTLAND

CANADIAN MILK NETWORK December 1962

Radiation Protection Division Department of National Health and Welfare, Ottawa, Canada

Dried Milk Products

Beginning in November 1955, radiochemical analyses of skim milk and buttermilk powders for strontium-90 concentrations were initiated by the Department of National Health and Welfare. Since April 1962, analyses for strontium-89 and cesium-137 have also been included.

Samples are collected through the cooperation of the Dairy Products Division of the Canadian Department of Agriculture, whose inspectors pick up four 1-pound samples of dried milk from each station (see figure 18) on a monthly schedule.

No statistical plan is followed for sample collection. Because of uncertainties introduced by this method of sampling, the significance of differences



FIGURE 18.—CANADIAN MILK SAMPLING STATIONS

between station-to-station and month-to-month results is not precisely known. However, it is possible to consider all results for a given period of time as being sufficiently random in selection to show any national trend when average values for such periods and all stations are plotted, as in figures 19 and 20.

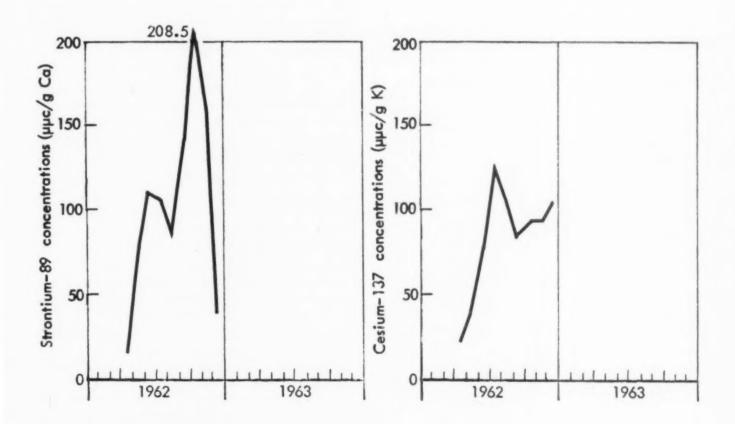


FIGURE 19.—AVERAGE STRONTIUM-89 AND CESIUM-137 CONCENTRATIONS IN MILK POWDER

A detailed discussion of the sampling and radiochemical procedures employed may be found in the Department's publications (1, 2). Table 10 presents the results of measurements of strontium-89. strontium-90, and cesium-137 in Canadian dried milk powder for December 1962. These data were taken from the "Data from Radiation Protection Programs," dated February 1963, published by the Radiation Protection Division of the Department of National Health and Welfare.

TABLE 10.—RADIONUCLIDES IN CANADIAN DRIED MILK POWDER, DECEMBER 1962

Station		Strontium-90 (μμc/g Ca)	Cesium-137 (μμc/g K)
Arborg	8	_	
Calgary	86.8	40.8	90
E. Florenceville		_	-
Edmonton		25.3	87
Grunthal		17.0	58
Halifax	18.9	19.0	103
La Durantaye	24.5	39.6	162
Lawrenceville	19.7	25.5	118
London		10.4	58
Megantic	15.6	40.3	163
Moneton	17.4	31.5	140
Nicolet		21.2	78
Ottawa		20.0	_
Saskatoon	102.4	18.9	59
Sussex		59.8	131
Vancouver	52.1	25.2	113
Walkerton	20.8	11.8	61
Average	39.2	27.1	103

a Dash indicates no sample.

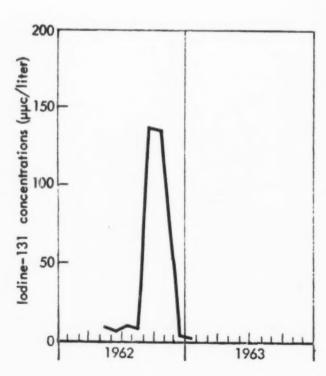


FIGURE 20.—IODINE-131 CONCENTRATIONS IN WHOLE MILK

During 1962, there has been almost a continual increase in the national average strontium-90 concentrations from the low of 6.9 $\mu\mu$ c/g calcium in March to 32.7 $\mu\mu$ c/g calcium in November (see figure 21). In December, the national average decreased to 27.1 $\mu\mu$ c/g calcium. However, the national average for 1962 was 19.4 $\mu\mu$ c/g calcium.

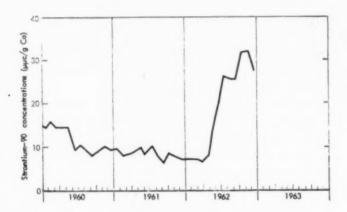


FIGURE 21.—AVERAGE STRONTIUM-90 CON-CENTRATIONS IN CANADIAN DRIED MILK POWDER

The national average strontium-89 concentration in powdered milk dropped to 27.1 $\mu\mu$ c/g calcium in December. As shown in figure 19 there is quite a pronounced peak in the national average strontium-89 concentrations in 1962.

For an approximate method to compare the Canadian powdered milk data in units of $\mu\mu c/g$ calcium to other milk data in units of $\mu\mu c/g$ calcium to other milk data in units of $\mu\mu c/g$ calcium/liter of liquid milk. Canadian powdered milk data may be compared to liquid milk data by multiplying the former values by 1.2.

Whole Milk

In April 1962, monitoring of liquid whole milk for iodine-131 was begun. Table 11 presents the December 1962 and January 1963 average concentrations of iodine-131 in liquid whole milk for nine cities. It may be noted that the national average which increased during the fall of 1962 has decreased to $7 \mu\mu c/liter$ in December 1962 and has remained low (3 $\mu\mu c/liter$) during January 1963 (see figure 20) indicating very little fresh fallout entering the milk.

It should be emphasized again that the interpretation of fallout data in relation to health is a complex problem. In comparing the values of the concentration levels in a particular medium

TABLE 11.—IODINE-131 IN CANADIAN LIQUID WHOLE MILK

[Concentrations in unc/liter]

Station	December 1962	January 1963
Calgary Halifax Ottawa Quebec Saskatoon Sault Ste. Marie Vancouver Windson Windson	9 (12) ^a 10 (10) 5 (10) 7 (11) 4 (10) 8 (11) 8 (10) 12 (7) 3 (12)	5 (13 3 (12 2 (13) 4 (13) 3 (12) 2 (13) 5 (13) 2 (13)
Average	7	3

^{*} Numbers in parentheses indicate the number of samples on which the monthly average is based.

with the so-called Maximum Permissible Concentrations (MPC's) as established by the International Commission on Radiological Protection

(3), it is necessary to keep in mind that the MPC values refer to conditions of continuous exposure over a lifetime. Therefore, the average levels over an extended period, such as one year, represent a better basis for comparison than do individual levels at any specific time.

REFERENCES

(1) Radiation Protection Division, Department of National Health and Welfare, Ottawa, Canada: The Preliminary Report on the Measurements of Radioactive Strontium in Canadian Milk Powder Samples, CNHW (RPD-1) (July 1958).

(2) Mar, Peter G.: Outline of Procedure for the Radiochemical Analysis of Dried Milk Powders for Strontium and Yttrium, (RPD-5), Radiation Protection Division, Department of National Health and Welfare (June 1, 1960).

(3) Recommendations of the International Commission on Radiological Protection: Report of Committee II on Permissible Dose for Internal Radiation, Pergamon Press, New York (1959).

MEXICAN MILK NETWORK November-December 1962

Radiological Protection Program of the National Commission of Nuclear Energy, Mexico

The Pasteurized Milk Monitoring Program of Mexico was established in March 1962 by the "Comision National de Energia Nuclear" through its Radiological Protection Program to provide a means for estimating the ingestion of strontium-90 from milk.

Milk has been analyzed from twelve cities distributed throughout Mexico. Since each city is supplied from a large number of sources, it has not been possible, to date, to apply a representative sampling method. However, samples from at least four main sources that supply large population groups were collected in each city.

The results presented in table 12 give an estimation of the strontium-90 content of the milk consumed in Mexico.

TABLE 12.—STRONTIUM-90 IN MEXICAN PASTEUR-IZED MILK, NOVEMBER-DECEMBER 1962

[Monthly averages]

Month (1962)	Sampling location	Calcium (g/liter)	Strontium-90 (µµc/liter)	Strontium-90 (µµc/g Ca)
November	México, D. F Puebla	1.07 1.13	2.9	2.7
December	Poza Rica México, D. F Casas Grandes	1.15 1.16 1.10	1.1 1.35 2.6	1.0
	Tuxtla Gutierrez . Villahermosa	1.12 1.38	1.0	1.0

a México City.

Twelve Month Sum of Daily Radionuclide Content of One Liter of Pasteurized Milk

Iodine-131: March 1962–February 1963 Strontium-89 and strontium-90: February 1962–January 1963

Division of Radiological Health, Public Health Service

The guidance of the Federal Radiation Council (FRC) is given in terms of transient rates of intake of radioactive materials in micromicrocuries per day. The action ranges as proposed in FRC Report No. 2 are based on radiation doses considered acceptable for lifetime exposure from normal peacetime atomic industry operations (1). The Council recommends the use of a time period of one year as an appropriate interval for averaging exposures and emphasizes that the annual acceptable exposure dose is not a "danger point" which, if exceeded, requires protective measures (1, 2, 3,).

To facilitate comparison of the concentrations of certain radionuclides in milk with the Radiation Protection Guides, tables 1 and 2 below furnish a means towards estimating the contribution of milk to the total dietary intake of iodine-131, strontium-89, and strontium-90. The tables are developed from the PHS Pasteurized Milk Network monthly averages of the radionuclides. They present index values which are the estimated sum of the daily amounts of a radionuclide in one liter of milk for a 12-month period.

The tables show 12-month index values for each of the Network's 62 sampling locations. Due to the longer time required for strontium-89 and strontium-90 analysis, these 12-month index values are for the year beginning one month earlier than the iodine-131 values. The columns of monthly index values in each table are used to compute the net change as the yearly index values are advanced by one month. The following column shows this new 12-month index value. In addition, the second column in table 1 gives the iodine-131 February 1963 concentration averages.

The data in tables 1 and 2 are calculated as follows: (a) results from all samples collected in each week (Sunday through Saturday) are averaged, (b) the weekly averages for all weeks ending within a given month are averaged and an average for the month is obtained, and (c) the monthly radionuclide index value is determined by multiplying the average for the month by the number of days

in the month. The number of days in the month will be either 28 or 35, corresponding to the complete calendar weeks used for any month. Procedures exemplified by (a) and (b) above minimize the effect of any one day's sample results on the average for the month, particularly for a short-lived radionuclide such as iodine-131. Yearly index values are obtained by the following procedure. In column (A) are the twelve-month index values for the period indicated. In columns (B) and (C) are the monthly index values for the periods indicated. The values in column (D) are obtained by adding the values in column (B) to those in column (A) and subtracting those in (C).

For a number of reasons it is desirable to have a standard quantity of milk to use in the development of index values for the different radionuclides. When one is concerned with strontium, 1 liter is a suitable quantity, as this amount of milk supplies approximately 1 gram of calcium, the amount used by the Federal Radiation Council in deriving the intake guidance for strontium. When one is concerned with iodine-131, the critical age group is the young infant. Available information suggests that the average milk consumption of infants in the 6-18 month group is not more than 1 liter per day. Thus the index value based on 1 liter of milk, though not directly an average intake value, is probably the most useful index for estimating total intake.

REFERENCES

 Federal Radiation Council: Background Material for the Development of Radiation Protection Standards, Report No. 2, Superintendent of Documents, U. S. Government Printing Office (September 1961), price 20 cents.

(2) Chadwick, Donald R., and Conrad P. Straub: Considerations in Establishing Radiation Protection standards for Radioactivity in the Environment, Radiological Health Data, 3:159-65, Superintendent of Documents, Government Printing Office, Washington 25, D.C. (May 1962).

(3) Public Health Service: Special Report, Radiological Health Data, 3:ii-iii, Superintendent of Documents, Government Printing Office, Washington 25, D.C.

(September 1962).

TABLE 1.—TWELVE-MONTH SUM OF DAILY AMOUNTS OF IODINE-131 IN ONE LITER OF MILK

[µµc day/liter]a

		Feb. 1963		Iodine-131	index value	
Sar	npling locations	iodine-131 averages (μμc/liter)	Feb 1962- Jan 1963 (A)	Feb 1963 (B)	Jan 27, 1963- Feb 23, 1963 (C)	Mar 1962- Feb 1963 (D)
Ala: Alaska: Ariz: Ark: Calif:	Montgomery Palmer Phoenix Little Roek Sacramento San Francisco	<10 <10 <10 20 <10 <10	6,590 38,220 4,270 14,250 5,050 4,940	140 140 140 140 140 140	140 140 140 560 140 140	6,590 38,220 4,270 14,670 5,050 4,940
Colo: Conn: Del: D. C: Fla: Ga:	Denver Hartford Wilmington Washington Tampa Atlanta	<10 <10 10 <10 20 10	5,860 7,670 11,730 8,440 6,240 8,900	280 140 140 140 140 140	140 140 280 140 560 280	5,720 7,670 11,870 8,440 6,660 9,040
Hawaii: Idaho: Ill: Ind: Iawa: Kans:	Honolulu	20 <10 <10 <10 <10 <10 <10	4,830 9,660 13,690 12,010 21,840 21,740	140 560 140 140 140 140	560 140 140 140 140 140	5,250 9,240 13,690 12,010 21,840 21,740
Ky: La: Maine: Md: Mass:	Louisville	<10 30 <10 <10 <10	10,540 8,900 8,160 8,690 7,950	140 140 140 140 140	140 840 140 140 140	10,540 9,600 9,160 8,690 7,950
Mich: Minn: Miss: Mo:	Detroit	$\begin{array}{c} < 10 \\ 10 \\ < 10 \\ < 20 \\ < 10 \\ < 10 \end{array}$	12,820 9,420 16,170 9,110 30,380 12,530	140 140 140 140 280 140	140 280 140 560 140 140	12,82(9,56(16,17(9,53(30,24(12,53(
Mont: Nebr: Nev: N. H: N. J: N. Mex:	HelenaOmahaLas VegasManchesterTrentonAlbuquerque	10 <10 <10 20 <10 <10	14,910 19,530 b4,100 7,290 7,990 7,460	560 280 	280 140 140 560 140 140	14,630 19,390 54,240 7,710 7,990 7,040
N. Y: N. C: N. Dak:	Buffalo	<10 <10 <10 <10 <10	8,720 11,660 9,980 3,370 14,910	140 140 140 140 140	140 140 140 140 140	8,72 11,66 9,98 3,37 14,91
Ohio: Okla: Ore: Pa:	Cincinnati	10 <10 10 <10 <10 <10	14,460 11,100 18,240 9,770 10,820 14,670	140 140 140 140 140 140	280 140 280 140 140 280	14,600 11,100 18,380 9,770 10,820 14,810
P. R: R. I: S. C: S. Dak: Tenn:	San Juan	30 10 20 10 <10 <10	45,260 8,440 6,590 14,530 7,850 10,050	140 140 140 140 140 140	840 280 560 280 140 140	45,96 8,58 7,01 14,67 7,85 10,05
Tex: Utah: Vt: Va:	Austin	<10 20 10 <10 <10	11,040 18,420 31,920 8,380 6,410	140 140 280 140 140	140 560 280 140 140	11,040 18,840 31,920 8,380 6,410
Wash: W. Va: Wis: Wyo:	Seattle	<10 10 10 10 <10	9,940 21,770 6,830 14,320 19,850	140 140 140 140 280	140 280 280 280 140	9,94 21,91 6,97 14,46 19,71

^a The data in this table are index values, not to be interpreted as consumption or total intake values. Annual iodine–131 intake per person may be calculated from an index value in this table by applying the appropriate factor representing average individual daily milk consumption for any selected group under consideration. Example: 12-month I ¹³¹ index \times milk consumption factor = 12-month I ¹³¹ intake ($\mu\mu c$ day/liter) (liter/day/person) (μc column μc

respectively.

^c A dash indicates no analysis.

^d No sample was received in November 1962. These sums are therefore for 11 months.

TABLE 2.—TWELVE-MONTH SUM OF DAILY AMOUNTS OF STRONTIUM-89 AND STRONTIUM-90 IN ONE LITER OF MILK

				index values y/liter)a			Strontium-96 (µµc da	0 index values y/liter)a	
San	npling locations	Jan. 1962- Dec. 1962 (A)	Jan. 1962 (B)	Dec. 30, 1962- Jan. 26, 1963 (C)	Feb. 1962- Jan. 1963	Jan. 1962- Dec. 1962 (A)	Jan. 1962 (B)	Dec. 30, 1962- Jan. 26, 1963 (C)	Feb. 1962- Jan. 1963 (D)
Ala: Alaska: Ariz: Ark: Calif:	Montgomery Palmer Phoenix Little Rock Sacramento San Francisco	21,385 18,665 6,570 48,215 6,905 11,610	70 270 810 1,350 270 270	2,240 700 700 3,500 420 1,680	23,555 19,095 6,460 50,365 7,055 13,020	5,324 3,607 1,145 10,661 1,363 1,895	270 135 81 567 54 54	420 336 112 812 84 140	5,474 3,808 1,176 10,906 1,393 1,981
Colo: Conn: Del: D. C: Fla: Ga:	Denver Hartford Wilmington Washington Tampa Atlanta	10,435 8,000 12,985 12,550 8,980 30,580	405 70 70 70 405 2,160	280 140 280 280 840 2,380	10,310 8,070 13,195 12,760 9,415 30,800	3,529 3,861 4,799 5,164 3,459 6,401	162 270 270 243 162 297	364 336 448 364 336 504	3,731 3,927 4,977 5,285 3,633 6,606
Hawaii: Idaho: Ill: Ind: Iowa: Kans:	Honolulu	9,870 8,730 10,800 13,720 23,670 18,185	945 135 70 70 270 405	1,120 280 70 280 560 840	10,045 8,875 10,800 13,930 23,960 18,620	1,858 3,230 3,886 4,772 4,461 4,096	108 108 162 243 135 162	168 364 392 420 392 336	1,918 3,486 4,116 4,949 4,718 4,270
Ky: La: Maine: Md: Mass:	Louisville New Orleans Portland Baltimore Boston	27,640 61,790 10,800 11,815 11,290	8,100 70 70 70	840 4,480 70 70 140	28,210 58,170 10,800 11,815 11,360	7,030 10,854 4,982 5,404 5,731	324 648 243 280 243	588 840 560 420 504	7,294 11,044 5,299 5,544 5,992
Mich: Minn: Miss: Mo:	Detroit	9,785 8,350 20,135 55,930 30,505 19,100	70 70 270 5,670 405 270	140 280 420 4,760 840 560	9,855 8,560 20,285 55,020 30,940 19,390	3,956 3,590 5,818 8,740 5,145 4,843	162 216 162 648 189 216	448 420 560 700 448 336	4,242 3,794 6,214 8,795 5,404 4,966
Mont: Nebr: Nev: N. H: N. J: N. Mex:	Helena Omaha Las Vegas Manchester Trenton Albuquerque	16,025 20,740 54,760 10,345 9,925 6,165	135 405 70 70 540	700 700 280 140 70 560	16,590 21,035 55,040 10,415 9,925 6,185	4,203 4,576 ⁵ 994 5,072 3,982 1,611	108 135 270 216 162	392 448 168 560 392 168	4.48 4.88 51.16 5.36 4.15
N. Y: N. C: N. Dak:	Buffalo New York Syracuse Charlotte Minot	8,770 10,555 9,555 20,360 16,370	70 70 70 270 135	280 70 70 700 420	8,980 10,555 9,555 20,790 16,655	4,105 4,951 3,942 6,856 6,699	297 324 162 297 189	420 448 448 700 784	4,22 5,07 4,22 7,25 7,29
Ohio: Okla: Ore: Pa:	Cincinnati Cleveland Oklahoma City Portland Philadelphia Pittsburgh	17,355 10,660 25,440 29,605 10,205 12,215	135 70 1,080 1,080 70	560 140 1,820 1,960 280 70	17,780 10,730 26,180 30,485 10,415 12,215	5,247 4,234 5,997 4,879 4,493 5,546	270 216 243 189 216 324	504 420 476 336 504 532	5,48 4,43 6,23 5,02 4,78 5,75
P. R: R. I: S. C: S. Dak: Tenn:	San Juan Providence Charleston Rapid City Chattanooga Memphis	b23,920 8,980 24,720 19,785 42,435 38,185	3,375 70 1,620 270 1,485 2,835	4,200 280 1,680 700 1,960 2,660	b24,745 9,190 24,780 20,215 42,910 38,010	b3,329 4,521 6,832 5,482 7,851 7,587	270 216 378 162 270 405	336 392 560 420 616 616	b3,39 4,69 7,01 5,74 8,19 7,79
Tex: Utah: Vt: Va:	Austin Dallas Salt Lake City Burlington Norfolk	27,715 10,935 10,970	1,080 135 135 270	1,960 3,640 420 140 560	11,655 30,275 11,220 10,975 17,395	2,558 5,305 3,166 4,066 6,345	108 216 135 216 297		2,70 5,59 3,36 4,29 6,55
Wash: W. Va: Wis: Wyo:	Seattle Spokane Charleston Milwaukee Laramie	21,505 13,105 20,785	540 135 135 70 135	420 280	21,945 13,530 21,070 8,455 19,165	5,433 4,453 6,591 3,046 3,552	162 162 270 162 108	364 532	5,52 4,65 6,85 3,13 3,86

a The data in this table are index values, not to be interpreted as consumption or total intake values. Annual strontium-89 or strontium-90 intake per person may be calculated from an index value in this table by applying the appropriate factor representing average daily milk consumption for any selected group under consideration.

Example: 12-month index value × milk consumption factor = 12-month intake $\frac{(\mu\mu\text{c day/liter})}{(\mu\mu\text{c day/liter})} \frac{(\mu\mu\text{c person})}{(\mu\mu\text{c person})}$ b Station included in milk network in July 1962. The sums in columns A and D are for 6 and 7 months, respectively.
c A dash indicates no analysis.
d No sample was received for November 1962. These sums are therefore fo 11 months.

SECTION III.—WATER

Radioactivity in Raw Surface Waters

NATIONAL WATER QUALITY NETWORK November 1962

Division of Water Supply and Pollution Control, Public Health Service

Radioactivity levels in the surface waters of the United States have been under surveillance by the Public Health Service through the National Water Quality Network since this nationwide sampling program was initiated in 1957. Beginning with the the establishment of 50 sampling points, this network has been expanded as of March 1963, to 125 stations (figure 1), which are operated jointly with State, Federal and local agencies and industry. The stations are located on the major waterways used for public water supplies, propagation of fish and wildlife, and recreational, agricultural, and industrial purposes. At these stations, samples are taken weekly, monthly, or continuously, depending on the type of analysis to be performed and on the water quality. The samples are then analyzed for plankton population, organic chemicals, chemical, biological, and physical quality, and radioactivity (1,3).

Radioactivity associated with dissolved solids

provides a rough measure of levels which may be found in treated water, since nearly all of the suspended matter is removed by treatment processes (4). It has been observed that in water the natural environmental beta activity is usually several times that of the natural environmental alpha activity. Nuclear installations may contribute additional alpha or beta activity whereas fallout contributes primarily additional activity. Gross alpha and beta measurements are made on both suspended and dissolved solids (strontium-90 on the total solids only) in raw surface water samples according to established procedures (5, 6).

For the first two years of the network's operations, beta determinations were made on weekly samples, and alpha determinations were generally made on composites of more than one weekly sample. From January 1960 to September 1961, alpha and beta determinations were generally made once a month on weekly composited samples.



Figure 1.—TOTAL BETA ACTIVITY (µµc/liter) IN SURFACE WATER AT NATIONAL WATER QUALITY NETWORK SAMPLING STATIONS, NOVEMBER 1962

Table 1.—RADIOACTIVITY IN RAW SURFACE WATERS, NOVEMBER 1962

[Average concentrations in pc/liter]

	November 1962						
Station		ta activi	ty	Alı	ha activ	ity	
		Dis- solved	Total	Sus- pended	Dis- solved	Total	
Allegheny River: Pittsburgh, Pa		_		_		-	
nimas River: Cedar Hill, N. Mex	283	37 12	320 15	9	3 0	1	
rkansas River: Coolidge, Kans	91	146	237	2	39	4	
Ponea City, Okla	32	39	71	1	3	7	
ear River; Preston, Idahoighorn River; Hardin, Mont	9 16	25 32	34 48	<1 2	6		
oux River: Sioux Falls, S. Dak	11	42	53	_	-	-	
hattahoochie River: Atlanta, Ga	9	17	26	1	0		
Columbus, Ga	17 22	18	35	1	0		
Lanett, Ala	10	12 21	34	1 0	0		
lear Water River: Lewiston, Idaholinch River:	12	14	26	0	0		
Kingston, Tenn Clinton, Tenn	-	_	_		-		
Clinton, Tennolorado River:	5	18	23	0	<1	<	
Loma, Colo	59	40	99	2	8	1	
Page, Ariz Boulder City, Nev	108	34 15	142	21	7 9	2	
Parker Dam, CalifAriz	-	-	_	_	-		
Yuma, Ariz	37	109	146	0	10	1	
Northport, Wash	5	18	23	0	<1	<	
Wenatchee, Wash. Pasco, Wash.	28	10 544	10 572	1 0	2		
McNary Dam, Ore	17	183	200	0	1		
Bonneville, Ore	22 101	122 81	144 182	0	1 0		
onnecticut River:							
Wilder, Vt	8 7	19	27 26	0	<1	<	
Enfield Dam, Conn	8	16	24	0	<1	<	
Delaware River: Martins Creek, Pa	7	23	30	0	0		
Trenton, N. J.	12	22	34	0	0		
Philadelphia, Pa scambia River: Century, Fla	16	24	40	0	0		
ireat Lakes:	-		10				
Duluth, MinnSault Ste. Marie, Mich	5 3	5 8	10	0	0		
Milwaukee, Wis	1	14	15	0	0		
Gary, Ind Port Huron, Mich		13 22	14 28	0	0		
Detroit, Mich	3 4	19	22 10	0	0		
Green River: Dutch John, Utah	27	29	56	0	2		
Iudson River: Poughkeepsie, N. Y	12	135	147	1	0		
Peoria III	5	24	29	1	1		
Grafton, III. Kanawha River: Winfield Dam, W. Va Klamath River: Keno, Ore Jitle Miami River: Cincinnati, Ohio	24 19	29 17	53 36	1	0		
Klamath River: Keno, Ore	8	18	26	0	0		
Attle Miami River: Cincinnati, Ohio	14	38 67	52 76	0	1 0		
Mississippi River:							
St. Paul, Minn Dubuque, Iowa	27	31 29	40 56	0	0		
Burlington, Iowa	310	24	334	0	0		
E. St. Louis, Ill. Cape Girardeau, Mo	20	37 48	46 68	1	1 2		
W. Memphis, Ark	56	28	84	0	2		
Delta, La New Orleans, La	10	31	41	1	3		
Vicksburg, Miss Missouri River:		33	58	2	1		
Williston, N. Dak	9	17	26	1	4		
Bismarck, N. Dak Yankton, S. Dak	2 3	18 50	20 53	1 0	4		
Omaha, Nebr	. 13	22	35	2	4		
St. Joseph, Mo	18 26	45 30	63	3 2	10		
St. Louis, Mo.	-	-	-	-	-		
Missouri City, Mo Monongahela River: Pittsburgh, Pa	. 38	44 22	82 25	7 0	9		
North Platte River: Henry, Nebr		45	55		23	1 3	
Ohio River: E. Liverpool, Ohio	1	13	14	0	0		
Addison, Ohio	. 6	26	32	<1	1		
Huntington, W. Va	41	31 26	72 41	2	0		
Louisville, Ky	34	24	58	1	0		
Evansville, Ind	34 21	34 19	68 40	1 6	0		
Cairo, Ill Ouchita River: Bastrop, La Pend Oreille River: Albeni Falls Dam, Idaho	13	16	20	0	1		
Platte River: Plattsmouth, Nebr.	23	36 41	49 64	0 2	<1 6	1	

TABLE 1.—RADIOACTIVITY IN RAW SURFACE WATERS, NOVEMBER 1962—Continued

[Average concentrations in pc/liter]

	November 1962						
Station	Be	ta activ	ity	Alpha activity			
	Sus- pended	Dis- solved	Total	Sus- pended	Dis- solved	Total	
Potomac River:							
Williamsport, Md	20	15	35	0	0		
Great Falls, Md	29	80	109	1	0		
Rainy River:	0	04	40				
Baudette, Minn	9	31 20	40	0	0		
Red River: Grand Forks, N. Dak	4	33	24	<1	0	<	
Red River, South:		33	34	_		-	
Denison, Tex	36	41	77	0	1		
Index, Ark	43	46	89	1	1		
Alexandria, La	122	31	153	4	Ô		
Bossier City, La	297	79	876	7	8	1	
Rio Grande River:			0.0				
Alamago, Colo	16	18	34	1	0		
El Paso, Tex	22	39	61	1	6		
Laredo, Tex	110	51	161	-	-	-	
Brownsville, Tex	6	16	22	1	5		
Roanoke River: John H. Kerr Resr. & Dam, Va	7	19	26	0	0		
Sabine River: Rubliff, Tex	_	_	_		-	-	
Sacramento River:							
Greens Landing, Courtland, Calif	13	15	28	1	1		
San Joaquin River: Vernalis, Calif	31	24	55	1	3		
San Juan River: Shiprock, N. Mex	95	51	146	11	6	1	
St. Lawrence River: Massena, N. Y Schuylkill River, Philadelphia, Pa	24 26	23	47	0	0		
Savanah River:	20	20	40		0		
North Augusta, Ga	3	8	11	0	0		
Port Wentworth, Ga	7	25	82	0	0		
Shenandoah River: Berryville, Va	14	30	44	<1	1		
Ship Creek: Anchorage, Alaska	0	7	7	0	0		
Snake River:							
Ice Harbor Dam, Wash	7	23	30	<1	2		
Wawawai, Wash	-	_	-	-	-	-	
Payette, Idaho	9	29	38	<1	2		
South Platte River: Julesburg, Colo	28	178	206	2	28	3	
Spokane River: Post Falls, Idaho	12	12	24	0	0		
Susquehanna River:	0.4	20	44	0	0		
Sayre, Pa	24 10	15	25	0	0		
Tenneasee River:	10	10	20	0	0		
Chattanooga, Tenn	16	54	70	0	0		
Bridgeport, Ala	2	27	29	0	1	1	
Bridgeport, Ala Pickwick Landing, Tenn	2	36	38	0	0		
Lenoir City, Tenn	9	24	33	<1	0	1 <	
Tombigbee River: Columbus, Miss	12	15	27	0	<1	1 3	
Truckee River: Farad, Calif	5	11	16	-	-	-	
Verdigris River: Nowata, Okla	10	47	57	<1	2	1	
Wabash River: New Harmony, Ind.	93	37	130	5	1	1	
Willamette River: Portland, Ore	39	15	54	<1	0	<	
Yakima River: Richland, Wash	3	25	28	0	1		
Yellowstone River: Sidney, Mont	5	35	10	0	3		

Beginning in September 1961, alpha determinations have been made on one sample each month, and beta determinations have generally been made on weekly samples. For the first operating year of each new station, sampling, and alpha and beta analysis are done weekly.

If at any time activity significantly greater than the normal environmental levels has been noted. the rate of sampling and analysis has been increased to at least one every week. Since January 1959, a portion of each sample from all stations in the network has been composited into a threemonth station sample for measurement of strontium-90 (7). Because strontium-90 analyses are done quarterly, the results will be published on this basis.

Table 1 presents the results of the alpha and beta analyses on raw surface water in the United States for November 1962. These data are preliminary; reanalysis of some samples may be made and additional analyses, not completed at the time of the report, may become available. For final data one should consult reference (8).

Quarterly strontium-90 results for the past year are shown in table 2. These data have been confirmed.

In order to obtain a geographical perspective of the radioactivity in surface water, the numbers alongside the various stations in figure 1 give the total beta activity in suspended-plus-dissolved solids in raw water collected at that station.

REFERENCES

- (1) Public Health Service: National Water Quality Control Network, Fallout from Nuclear Weapons Tests, 1:167-9, Hearings before the Special Subcommittee on Radiation of the Committee on Atomic Energy, Superintendent of Documents, U.S. Government Printing Office, Washington 25, D.C. (May 1959), price \$2.75.

 (2) Division of Water Supply and Pollution Control, Public

TABLE 2. -QUARTERLY STATION AVERAGE CONCENTRATIONS OF STRONTIUM-90 IN RAW SURFACE WATERS, JULY 1961-JUNE 1962

[Concentrations in pc/liter]

Station	Third quarter 1961	Fourth quarter 1961	First quarter 1962	Second quarter 1962
Allegheny River: Pittsburgh, Pa Animas River, Cedar Hill, N. Mex.	0.8	0.6	1.7	1.2 1.0
Apalachicola River: Chattahoochie, FlaArkansas River:	0.4	0.5	*0.9	0.9
Coolidge, Kans	2.3	1.0	0.6	3.9
Pendelton Ferry, Ark Bighorn River: Hardin, Mont	=	0.5	1.3 *1.5	6.4
Big Sioux River: Sioux Falls, S. Dak	0.4	1.2	2.0	5.9
Chattahoochie River: Atlanta, Ga	-	0.3	0.5	0.8
Chena Slough: Fairbanks, Alaksa	_	0.1	0.6	1.4
Clear Water River: Lewiston, Idaho Clinch River: Kingston, Tenn	=	0.3	0.5	*0.4 1.7
Colorado River: Loma, Colo	_	0.8	1.2	1.5 4.9
Page, Ariz Boulder City, Nev Parker Dam, Calif-Ariz Yuma, Ariz	1.0	1.8 1.5 0.7	1.3 1.0 0.6	2.0 1.7 1.0
Columbia River: Northport, Wash	_	_		0.6
Northport, Wash Wenatchee, Wash Pasca, Wash	1.1	0.7	1.1	0.8
McNary Dam, Ore Bonneville Dam, Ore Clatskanie, Ore Connecticut River:	*0.6	0.6 0.8 0.6	0.9 1.1 0.7	0.7 1.6 0.7
Wilder, Vt	_	3.7 0.4	0.8 0.6	0.8
Clarksville, Tenn	0.4	0.9	_	_
Delaware River: Martins Creek, Pa Trenton, N. J. Philadelphia, Pa	=	0.5 0.7	0.7 1.0 *0.7	1.1 1.6 1.3
Escambia River, Century, Fla Great Lakes:	*0.9	0.5	0.4	1.1
Duluth, Minn. Sault Ste. Marie, Mich Milwaukee, Wis.	_	0.1 0.5 0.8 0.1	0.8 0.5 0.5 0.9	0.5 0.6 0.7 0.7
Gary, Ind. Port Huron, Mich. Detroit, Mich.	0.4	0.8	0.8	1.8
Hudson River: Poughkeepsie, N. Y.	0.2	0.8	1.0	
Illinois River: Peoria, Ill Grafton, Ill Kanawha River:	0,4	0.7	1.0 2.5	1.7
Winfield Dam, W. Va	=	0.5	0.4	0.8
Cincinnati Ohio	1.1		1.3	2.9
Merrimack River: Lowell, Mass Mississippi River: St. Paul, Minn	0.9			
Dubuque, Iowa Burlington, Iowa	0.6	0.8 1.9 0.9	1.0	2.1
E. St. Louis, Ill	0.8		1.4	2.1
W. Memphis, Ark Delta, La New Orleans, La Vicksburg, Miss	*0.4		1.0	1
Missouri River: Williston N. Dak		1.2	1.4	1.
Bismarck, N. Dak. Yankton, S. Dak. Omaha, Nebr.	0.6	2.5	0.7	2. 4.
St. Joseph, Mo Kansas City, Kans Missouri City, Mo		0.5	0.5	1.
St. Louis, Mo	1.4		1.3	2.
Pittsburgh, Pa. North Platte River: Henry, Nebr. Ohio River:		1.5	0.5	2.
Ohio River: E. Liverpool, Ohio. Huntington, W. Va. Cincinnati, Ohio Louisville, Ky.	0.4	1.0		1.
Evansville, Ind	-	0.1	7 1.8	1.
Cairo, Ill Ouachita River: Bastrop, La Pend Oreille River:	-	0.		
Albeni Falls Dam, Idaho		0.	4 2.5	1 6

Station	Third quarter 1961	Fourth quarter 1961	First quarter 1962	Second quarter 1962
Potomac River: Williamsport, Md				
Great Falls, Md.	_	0.9	0.5	0.9
Rainy River:		0.0	1.1	2.2
International Fls., Minn	-	0.3	2.4	1.4
Baudette, Minn Red River, North:	_	-	1.4	2.4
Red River, North:				
Grand Forks, N. Dak Red River, South:	-	1.1	_	_
Denison, Tex		2.2	2.2	2.2
Index, Ark	_	1.7	2.1	6.1
Alexandria	1.0	2.7	1.1	2.5
Rio Grando Rivor				2.0
Alamosa, Colo El Paso, Tex Laredo, Tex Brownsville, Tex	*0.4	_	*0.6	1.1
El Paso, Tex		_	*0.5	1.2
Laredo, Tex	_	0.6	0.4	2.0
Roanoke River:	_	0.6	0.9	1.2
John H Kerr Rear & Dam Va		0.6	0.6	2.5
Sabine River: Ruliff, Tex	_	0.7	1.3	1.7
Sacramento River: Greens		1	1.0	1.1
Landing, Courtland, Calif	_	-	-	0.9
San Juan River:				1
Shiprock, N. Mex	_	1.3	1.9	1.0
St. Lawrence River: Massena, N. Y.				
Massena, N. Y		1.6	0.8	1.2
North Augusta, S. C.			0.8	1.0
Port Wentworth, Ga	0.4	0.6	0.9	1.9
Schuylkill River: Philadelphia, Pa	0.4	0.0	*0.6	1.5
Schuylkill River: Philadelphia, Pa- Shenandoah River: Berryville, Va-	-	0.2	0.4	0.9
Ship Creek: Anchorage, Alaska	-	-	_	*0.2
Snake River:				
Wawawai, Wash	0.3	0.2	0.9	0.6
Ice Harbor Dam, Wash Payette, Idaho	-	2.5	0.3	0.9
South Platte River:	_	2.5	0.3	0.6
Julesburg, Colo	.07	1.2	0.8	1.7
Spokane River: Post Falls, Idaho	-	_		0.7
Susquehanna River:	1			
Sayre, Pa	0.3	0.2	0.9	1.2
Conowingo, Md	0.3	0.9	0.5	1.0
Tennessee River:	0.0	0 -	0.0	
Chattanooga, Tenn	0.6	0.5	0.6	1.6
Bridgeport, Ala Pickwick Landing, Tenn	0.1		1.3	1.
Lenoir City, Tenn		-	1.2	1.3
Lenoir City, Tenn Tombigbee River: Columbus, Miss_	_	0.2	0.9	1.
Truckee River: Farad, Calif		0.1	0.8	0.5
Verdigris River: Nowata, Okla Wabash River: New Harmony, Ind.	_			3.
Wabash River: New Harmony, Ind.			1.3	4.
Yakima River: Richland, Wash		0.7		0.4
Yellowstone River: Sidney, Mont	_	1.2	1.3	30

^{*} Six months composite—added period required to obtain sufficient sam-

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(3) Setter, L. R., and S. L. Baker: Radioactivity of Surface
Waters in the United States, Radiological Health Data, 1:20-31 (October 1960).

(4) Straub, C. P.: Significance of Radioactivity Data, Journal of the American Water Works Association, 53:704 (June 1961).

(5) Setter, L. R., J. E. Regnier, and E. A. Diephaus: Radio-activity of Surface Waters in the United States, Journal of the American Water Works Association, 51:1377 (November 1959).

(6) Robert A. Taft Sanitary Engineering Center, Public Health Service: Radionuclide Analysis of Environmental Samples, Technical Report, R59-6 (1959).
(7) Straub, C. P., L. R. Setter, A Goldin and P. F. Hallbach: Strontium-90 in Surface Waters in the U.S., Journal of the Computer of

the American Water Works Association, 52:756 (June

(8) Division of Water Supply and Pollution Control, Public Health Service: National Water Quality Network Annual Compilation of Data, PHS Publication No. 663, 1962-1963 Edition, Superintendent of Documents, U.S. Government Printing Office, Washington 25, D.C. (To be published).

NORTH CAROLINA SURFACE WATER¹ June 1958–December 1961

Sanitary Engineering Division North Carolina State Board of Health²

The greater use of radioactive materials within the State of North Carolina presents an ever increasing concern to the State Board of Health and water works officials responsible for water quality. In view of this mounting problem, the Sanitary Engineering Division and the State Laboratory of Hygiene of the State Board of Health began in June 1958, with the cooperation of municipalities, a program of measurement of radiation background within the surface waters used as sources of public water supplies. Some 147 sampling points were established to sample raw surface water.

Background Radioactivity in Water

All waters contain traces of radioactivity which originate from naturally radioactive minerals dissolved from rock strata or from radioactive particulate material or gases in the atmosphere. Common among these minerals are trace elements of potassium-40, radium, thorium, and uranium. Such trace elements are dissolved by water as it finds its way to the watercourses, or as it flows within the watercourses. Precipitation is the major mechanism by which particulate matter or radio-

active gases such as thoron and radon, are removed from the atmosphere.

The combination of these radioactive materials imparts radioactive characteristics to the water and constitutes what is known as "background radioactivity" of the water.

A knowledge of the concentration of this background radioactivity is an important factor in the present and future appraisal of water quality for several reasons. First, once the background radioactivity has been changed due to the addition of isotopes, fission products, or other types of radioactive material, it may not be possible to establish what the original background may have been. Second, standards pertaining to radiation exposure or concentration within drinking water are expressed in terms of "additions to the natural background" (2). Third, the knowledge of background radioactivity is the future yardstick by which the extent of pollution of added radioactivity and the control of such material on the watershed areas can be determined. The present program was established to determine this baseline information.

Table 3 gives the average background radioactivity as well as the maximum and minimum values recorded over the period June 1958 through October 1961. The first column consists of code numbers which indicate the geographical positions of the sampling stations shown in figure 1.

Laboratory Procedure

Total solids content is determined for each

¹ Abstracted from reference (1) ² Raleigh, North Carolina

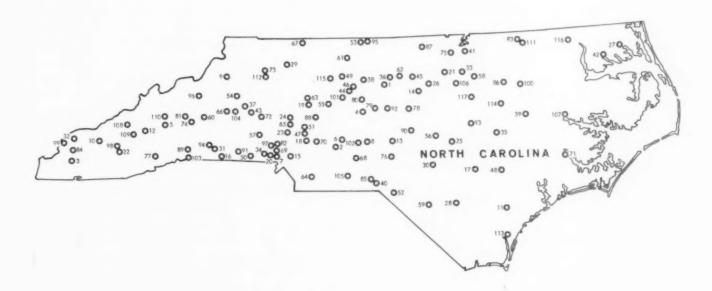


FIGURE 2.—NORTH CAROLINA SURFACE WATER SAMPLING STATIONS WITH CODE NUMBERS, 1958-1961

TABLE 3.—AVERAGE BACKGROUND RADIOACTIVITY FOUND IN RAW SURFACE WATER USED AS SOURCES OF PUBLIC WATER SUPPLIES, NORTH CAROLINA, JUNE 1958-OCTOBER 1961

Cont	Y	Source of water supply	Number	Background activity in µµc/liter		
Code umber	Location	Location Source of water supply		Average	Maxi- muma	Mini- mum
1 2 3 4 5 6 7 8 9	Alamance Albemarle Andrews Asheboro Asheville Badin Belmont Biscoe Blowing Rock Bryson City	Big Alamance Creek Yadkin River Beaver Creek Back Creek Bee Tree Creek Yadkin River Catawba River Little River Brickhouse Creek Lands Creek	16 29 16 15 35 16 15 14 33	7.04 5.11 2.58 7.94 2.17 6.50 6.14 6.34 4.92 1.89	9.77 17.39 6.11 32.62 8.19 14.11 23.77 7.72 14.69 5.90	3.8 1.9 1.1 2.8 0.8 2.8 1.8 3.2 1.0
11 12 13 14 15 16 17 18 19 20	Burgaw Canton Carthage Chapel Hill Charlotte Cliffside Clinton Concord Cooleemee Cramerton	NE Cape Fear River Pigeon River; Ruff Creek Spring Price Creek Catawba River Second Broad River Black River Cold Water Creek & Lumby Creek South Fork Yadkin South Fork Catawba River	18 15 16 16 33 33 29 34 16	9.18 1.63 4.18 5.39 4.61 8.70 5.77 10.63 7.66 8.46	29.51 3.51 9.87 15.9 18.08 36.12 31.12 38.05 16.70 21.20	2. 1. 1. 1. 1. 3.
21 22 23 24 25 26 27 28 29 30	Creedmoor Cullowhee Davidson Denton Dunn Durham Elizabeth City Elizabethtown Elkin Fayetteville	Ledge Creek Long Branch; Flat Creek Cathy's Creek Lick Branch Cape Fear River Flat River Pasquotank River Cape Fear River Big Elkin Creek Cape Fear River	16 15 14 15 11 32 16 21 17 30	10.60 2.37 4.80 13.04 15.97 9.71 15.96 6.84 3.98 6.94	17.8 4.86 6.9 31.66 49.90 25.97 23.6 24.52 12.15 39.84	2. 1. 2. 2. 5. 2. 10. 2.
31 32 33 34 35 36 37 38 39 40	Forest City Fontana Dam Franklinton Gastonia Goldsboro Graham Granite Falls Greensboro Greensboro Hamlet	Second Broad River Little Tennessee River Kearney's Creek Long Creek; Catawba River Little River; Neuse River Back Creek Catawba River Reedy Fork Creek Tar River Marks Creek	16 16 17 33 20 16 15 34 31	7.26 2.94 5.90 5.74 10.25 8.56 3.95 8.59 8.58 7.54	34.20 9.3 14.21 28.38 27.02 12.70 11.22 30.60 29.66 28.76	2 1 1 1 1 1 4 1 2 2 1
41 42 43 44 45 46 47 48 49 50	Henderson Hertford Hickory High Point Hillsboro Jamestown Kannapolis Kenansville Kernersville Kings Mountain	Sandy Creek Perquimans River Catawba River Deep River Eno River Deep River Buffalo Creek NE Cape Fear River Belew's Creek Kings Creek	16 30 27 17 13 33 11	5.91 15.30 4.41 11.52 6.84 15.39 7.07 11.94 6.81 2.35	18.69 26.10 12.25 35.26 21.4 26.86 29.72 47.00 19.15 19.63	2 5 1 2 2 1 1 1 1 2 1
51 52 53 54 55 56 57 58 59 60	Landis Laurinburg Leaksville Lenoir Lexington Lillington Lincolnton Louisburg Lumberton Marion	Grant's Creek Jordan Creek Dan River Zacks Fork Creek & Catawba River Abbott's Creek Cape Fear River Walker Branch Tar River Lumber River Mackey's Creek; Buck Creek	16 17 32 15 16 16 16 14 32	7.25 4.30 9.64 5.62 8.96 10.79 4.89 6.28 6.16 1.34	33.35 17.20 63.10 12.36 20.56 30.4 4.25 23.39 17.12 2.03	2 0 2 1 1 2 4 1 1 2 1
61 62 63 64 65 66 67 68 69 70	Mayodan Mebane Mocksville Monroe Mooresville Morganton Mount Airy Mount Gilead Mount Holly Mount Pleasant	Mayo River Mills Creek Bear Creek Richardson Creek Byers Creek; Catawba River Henry River Lovell's Creek Pee Dee River Catawba River Dutch Buffalo Creek	15 17 17 19 17 17	7.81 4.48 15.06 13.19 8.44 1.82 2.38 7.48 4.15 9.19	37.70 7.20 81.00 52.05 43.33 3.70 5.90 41.13 13.93 20.57	1 1 2 2 2 2 1 1 1 2 1 3
71 72 73 74 75 76 77 78 79 80	New Bern Newton North Wilkesboro Old Fort Oxford Finehurst Pisgah Forest Pittsboro Ramseur Randleman	Neuse River Jacobs Fork Reddies River Jarrett Creek Tar River: Hatcher's Run Rattlesnake Branch; Juniper Branch Davidson River Robinson Creek Sandy Creek Pole Cat Creek	11 31 16 16 15 15	5.07 9.01 6.03 2.04 9.30 3.59 3.59 8.98 6.28 15.95	10.34 28.49 52.72 5.23 43.88 12.85 12.85 35.89 14.51 69.31	1 1 1 1 1 1 0 0 3 2 3
81 82 83 84	Ridgecrest Riverbend Roanoke Rapids Robbinsville	Mountain Stream Catawba River Roanoke River Rock & Burgin's Creek	17 32	1.55 4.72 6.14 2.06	2.99 18.34 16.54 6.37	1 1 0

TABLE 3.—AVERAGE BACKGROUND RADIOACTIVITY FOUND IN RAW SURFACE WATER USED AS SOURCES OF PUBLIC WATER SUPPLIES, NORTH CAROLINA, JUNE 1958-OCTOBER 1961—Continued

			Number	Background activity in μμε/liter			
Code number		Source of water supply	of samples surveyed	Average	Maxi- muma	Mini- mumb	
85 86 87 88 89	Rockingham Rocky Mount Roxboro Salisbury Saluda Sanford	Falling Creek. Tar River Lake Issac Walton; Storey's Creek. Yadkin River Paces Creek Lick Creek; Potterage Creek.	11 15 17 14 17	6.27 13.17 10.55 12.31 1.80 8.67	18.42 57.10 35.74 60.20 3.50 47.35	1.33 3.31 2.98 2.57 1.06	
91 92 93 94 95 96 97 98 99	Shelby Siler City Smithfield Spindale-Rutherfordton Spray Spruce Pine Stanley Sylva Tapoco Tarboro	First Broad River Rocky River Neuse River Hollands Creek; Cathy's Creek Smith River Beaver Creek Hoyle's Creek Dill's Creek Yellow Hammer Creek Tar River	30 16 29 17 34 16 17 31	3.27 9.56 13.17 3.35 7.66 2.00 6.36 3.31 1.78 8.34	32.92 34.36 28.46 11.55 61.80 4.23 31.13 17.01 3.77 32.46	1.12 2.5 4.00 1.90 2.00 1.2 2.00 0.8 0.90	
101 102 103 104 105 106 107 108 109 110	Thomasville Troy Tryon Valdese Wadesboro Wake Forest Washington Waterville Waynesville Weaverville	Abbott's Creek Denson Creek; Downing Creek Falls Creek; Vaughn's Creek Micol Creek; Hoyle Creek; Catawba River Jones Creek Smith's Creek Tranters Creek Big Creek Cherry Cove & Shiney Creek Wagner Branch & Ox Creek	17 17 17 17 17 17	10.79 5.83 3.30 5.86 7.66 4.28 8.65 1.55 2.59	59.02 15.82 4.08 18.40 33.29 8.25 36.42 3.03 9.14 3.47	3.8 1.5 1.3 1.8 1.8 1.3 2.3 0.9 1.1	
111 112 113 114 115 115 116 117	Wedlon Wilkesboro Wilmington Wilson Winston-Salem Winston-Salem Winton Zebulon	Roanoke River Cub Creek Cape Fear River Contentney Creek Salem Creek Yadkin River Chowan River Little River	16 32 33 33 33 29	5.30 3.16 11.20 3.90 6.64 7.31 7.77 4.62	16.11 7.38 86.00 28.87 20.12 26.04 25.46 10.60	2.6 1.1 2.1 2.0 1.6 1.2 1.7	

Maximum values have 95 percent accuracy ± 20 percent.
 Minimum values have 95 percent accuracy ± 60 percent.

sample so that proper sample proportions may be used in analysis. The calculated amount of a sample, based on solids content, is concentrated to a few milliliters on a steam bath. The residue is transfered to a stainless steel planchet and evaporated to dryness under an infrared light. After further drying for one hour at 103°C., the sample is cooled in a dessicator and is then counted.

Alpha plus beta counting is done in a gas-flow internal proportional counter which is calibrated with a radium-D E source. On the vasis of instrument background, each sample is counted for a preset count and the results are reported in terms of micromicrocuries per liter.

Interpretation of Results

The measurement of radioactivity at low levels or low concentrations is difficult because of the many variable factors such as types and energy of radiation, necessary concentration to obtain a representative sample, and efficiency of counting which includes absorption, scatter, and geometry.

All results may be interpreted on the basis of the maximum permissible continuous occupational

concentration or radioactivity concentration guide set forth by the Subcommittee on Permissible Internal Dose of the National Committee on Radiation Protection. This has been established at 100 micromicrocuries per liter of water for the total activity of unknown isotopes (3). This value relates to average intake from water over a long period of time and not from transient conditions.

Individual sample data and associated errors are contained in reference (1). Only significant values were used in calculating the arithmetic averages given in this report. The significant increase in activity recorded for the period from November 9, 1961 to December 31, 1961 is probably due to the presence of fission material produced during the renewed atmospheric testing of nuclear weapons. For this reason, the values recorded during this period were not included in the calculations of the average background radioactivity.

From a study of the results, it can be noted that the mountain waters contain less radioactivity than those in the Piedmont and further east (see figure 1 and table 3). This is to be expected since the waters in the mountains contain less mineral content. As both surface and subsurface waters flow toward the ocean their dissolved solids content, paralleled by their concentrations of natural radioactivity, increases. Hence, from results to date, it cannot be assumed that the minerals in the Piedmont in the eastern part of the State necessarily contain more radioactivity than those in the western part.

Previous coverage in Radiological Health Data:

Period June-August 1962 (cistern water) Issue

February 1962

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SECTION IV.—OTHER DATA

Control of Luminous Dial Watches in New York City

Ira R. Paul¹

Introduction

The Office of Radiation Control of the New York City Department of Health was created for the purpose of administering a control program to protect the general public, as well as workers in certain installations, from the dangers inherent in the uncontrolled use of radioactive materials or ionizing radiation. Most of its efforts concern the use of X rays and radioactive materials, including radium, in medical installations, schools and institutions, but occasionally it is necessary to deal with other radiation problems. One of these, the use of radioactive materials in watches, is of particular interest because of the widespread use and popularity with young people of luminous-dial watches.

New York City's efforts to limit the public exposure from radium-dial timepieces began on March 19, 1958, with the enactment of Article 175 of the New York City Health Code which prohibited the "storage, manufacture, repair, handling, or use of any timepieces, instruments, novelties or devices, if done in such a manner that any person is exposed to radiation dose rates in excess of limitations recommended by the National Committee on Radiation Protection and Measurements (1)."

Upon the recommendation of the Mayor's

Technical Advisory Committee on Radiation, these dose rates were interpreted to limit the radiation emission through the front surface of pocket watches to no more than 0.1 mrad per hour and the radiation emission through the back surface of wrist watches to no more than 1.0 mrad per hour.

It was estimated, by general observation, that pocket watches are worn for an average of 100 hours per week, 50 weeks per year, facing the body in a pocket near the belt-line. Wrist watches may be worn with the back surface in contact with the skin of the forearm for approximately 150 hours per week, and thus it was necessary to establish separate criteria for each of these two styles of watches.

The National Committee on Radiation Protection and Measurements (NCRP) (2) has recommended that the maximum permissible dose (MPD) to the head, trunk, active blood-forming organs, gonads or lens of the eye of radiation workers shall not exceed 5 rems per year, and the maximum permissible dose to the hand or forearms shall not exceed 75 rems per year. The NCRP has further advocated the principle that the general population should not receive radiation at a weekly rate higher than one-tenth the respective permissible weekly dose for the critical organs and other body tissues.

The Federal Radiation Council's Radiation Protection Guides (RPG) (3) also recommended the above values not be exceeded for radiation workers. When considering individual members of

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the population, a whole body dose of 0.5 rem per year should not be exceeded. Similarly, with an average mumber of the population, a 30-year gonadal dose of 5 rem should not be exceeded. As an operational technique, where the individual whole body doses are not known, a suitable sample of the exposed population should be developed whose protection guide for annual whole body dose will be 0.17 rem per capita per year. The FRC has not recommended specific guides for exposure of the hands and forearms.

Using the NCRP dose rates as a guide for control at the radiation source rather than at the organ of interest, it then follows that if a pocket watch is worn for 5,000 hours per year, and the MPD is 0.5 rems (500 mrems), then the emission rate at the surface of the crystal should be a maximum of 0.1 mrad per hour. If a wrist watch is used for 7,500 hours per year, and the MPD is 7.5 rems (7,500 mrems), then the emission rate at the surface of the back should be a maximum of 1.0 mrad per hour.

Numerous wrist watches were tested and, with the exception of some skin-divers watches, were found to fall within the established limits.

Literature Survey

Joyet (4) reported in 1958 that the average man's luminous wrist watch contains 0.36 microcuries of radium and the average woman's wrist watch contains 0.13 microcuries. He calculated that a man wearing such a watch 24 hours a day receives a gondal dose of 21.8 mr per year and a woman receives a gondal dose of 12.7 mr per year.

Using Joyet's results, a sampling by the New York AEC operations office of 224 persons (including those not wearing watches and those wearing watches with non-luminous dials) in New York City indicated an average gonadal exposure of 3 mr per year per person to age 35. The 35-year dose is then about 0.1 r, which amounts to approximately 3 percent of natural background radiation and only about 1.5 percent of the total radiation dose derived from background plus medical and dental exposure to the gonads, as currently estimated (5).

Eng, LeCoultre and Lerch (6) examined 684 wrist watches produced in Switzerland in 1961, and found the highest percentage of these watches contained approximately 0.1 microcuries of radium-226 and emitted less than 1 mrem per hour through the back. Quite evidently, the trend

has been to reduce the amount of radium on watches.

Seelentag and Schmier (7) state that most German-made watches have radium amounting to less than 0.06 microcuries and usually only about 0.02 microcuries. The radiation exposure to wearers of such watches amounts to a small percentage (about 1 percent) of natural background radiation exposure, and the local exposure to the skin of the forearm amounts to only a few percent of the maximum permissible occupational dose.

Although no comparable study has been made with radium-dial pocket watches, the fact that they are worn in relatively close proximity to the gonads, roughly 28 cm for the average male, makes the pocket watch potentially a much greater genetic hazard than a wrist watch with equal amounts of radium-226. In considering the dose to the abdomen, the fact that the radium emanations are shielded only by the thin glass crystal and a few mg/cm² of cloth, rather than the mechanism and back of the wrist watch, again makes the pocket watch a potentially greater hazard to the wearer.

Survey Results and Laboratory Calibration Procedures

In 1960, a large department store and a large cutlery chain, both in New York City, advertised radium-dial pocket alarm watches. Samples of these watches were obtained and tested, and were found to emit 4 mrads per hour at their faces. The stores were notified, and they voluntarily removed these watches from sale.

In 1962, the manager of a mail order company, after being informed that the radiation emission from the pocket watches he had advertised was excessive, complained that there were many other sales outlets for these watches throughout the city. An inspection team was sent out and took a random sampling of 77 jewelry, department, novelty, and cigar stores. About half of these stores were found to be stocking radium dial pocket watches.

Rough field measurements, made with a portable end-window G-M survey meter having scale ranges of 0–30 and 0–100 mr/hr and a window thickness of 1.5 to 2.0 mg/cm², gave readings of from 3 to 10 mr/hr for various makes and models, with the window of the probe in contact with the surface of the watch crystal.

Admittedly, this procedure was crude, but since

all readings were on the low side due to poor geometry, it established the fact that all of the pocket watches checked exceeded the acceptable limits. Calibration of the instrument was obtained utilizing a 10.08 mg radium needle encapsulated in 0.5 mm of platinum, and using a factor of 0.825 to convert the mg of radium to mr per hour at a meter.

Two pocket alarm watches were placed in the whole body counter at the Atomic Energy Commission's New York Operations Office. Watch "A" was estimated to contain 0.15 microgram of radium, and watch "B" 0.015 microgram of radium. When checked with the G-M counter, the dose reading at the surface of the crystal of watch "A" was 8 mrad per hour and that of watch "B" was 1.2 mrad per hour.

To check depth of penetration in tissue, a pocket watch "C" which gave a reading of 8 mrad per hour on the G-M meter, and another pocket watch "D" which gave a reading of 3 mrad per hour, were used as radiation sources with an approximately tissue-equivalent pressed-wood phantom. Watch "C" had a gamma component that penetrated 8.25 inches of the phantom before reaching background levels and the gamma component of watch "D" penetrated 2.44 inches.

Again it is emphasized that the G-M meter is used only for fast field determinations. A more precise method, used in the laboratory, utilizes films. Exposure dose rates obtained by the film method are about three times higher than those obtained for the same watch by the G-M meter method.

Because a fast film is desirable, it was found convenient to use the same personnel dosimeter packet, Dupont type 556, that is used to monitor our laboratory and field personnel. This is a double film packet with a sensitive component, type 508, and an insensitive component, type 834. Only the sensitive film, the 508, which is at the front of the packet, is utilized for watch dosimetry.

Autoradiographs are obtained by taping two packets, side by side, on the face of the pocket watch, with the front of the packet in contact with the crystal. The films are left in place for two to three days. The time necessary to produce a satisfactory autoradiograph was previously determined by G-M counter measurement. During this time. the watch is kept wound, to simulate actual wearing conditions. In this way, the radium on the moving hands will distribute its energy over the area of the face, rather than concentrate it in one spot. Numbers or spots on the dial that are coated with radium are naturally stationary, and the film will receive greater exposure at these points. This is not too distinct, however, since a great deal of scattering takes place within the glass crystal, resulting in a blurred image. Figure 1 shows an example of an autoradiograph produced using this procedure. An autoradiograph made with the hands stationary is also shown (figure 2) to demonstrate that improper technique (watch not wound) would give false readings.

The films are developed along with unexposed films from the same emulsion batch. The purpose of these unexposed films is to establish a base and fog limit. After thorough drying, the net density of the films is determined with a densitometer. The "net density" is the density above base density and fog. Thus, net density is the density produced by the exposure (8).

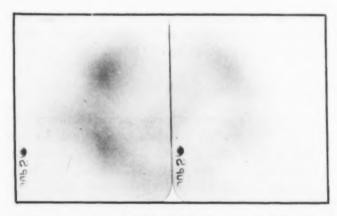


FIGURE 1.—ENLARGED PRINT OF AUTORADIO-GRAPH FOR WATCH WOUND DURING EXPOSURE TIME

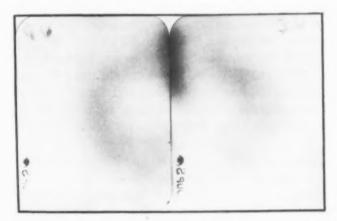


FIGURE 2.—ENLARGEMENT PRINT OF AUTO-RADIOGRAPH FOR WATCH UNWOUND DURING EXPOSURE TIME

The results are then compared against a radium-226 calibration curve (see figure 3). To establish this curve, the same radium source is used that is used to calibrate the G-M counter. Dupont type 556 film packets are placed a distance of one meter from the source, and removed individually at varying time intrvals. The exposure received by each film is checked with a Victoreen Condenser R-Meter thimble chamber. After development and determination of the net density of each film the net densities are plotted against exposure in milliroentgens on semilogarithmic paper, and a characteristic curve is obtained (9). Because the steepest slope occurs between net densities of 0.3 and 1.0, the exposure time of the watch film is estimated to bring the exposure into this curve segment.

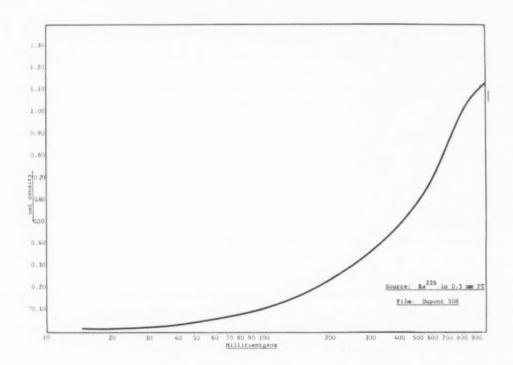


FIGURE 3.—RADIUM-226 CALIBRATION CURVE

This is, of course, a gamma curve to which a beta-gamma exposure is being related. The beta component of the watch-dial radium which penetrates the glass crystal, the white and green protective film papers (weight = 27.4 mg/cm²) and the film emulsion itself is in the energy range above 0.5 Mev (10). The response of photographic film has been studied by Fleeman and Frantz (11) for beta radiation energies ranging from 0.5 to 1.4 Mev. Within these limits, film response was found to be proportional to the dose received and independent of the energy of the incident electrons.

Most of the beta particles are filtered out before reaching the emulsion, but radium E (bismuth-210) has a 1.17 Mev beta which has a range of 500 mg/cm² (10). The beta-ray sensitivity of the film is almost proportional to the gamma-ray sensitivity,

and therefore the film densities are based upon a 1-to-1 beta-gamma ratio (12, 13).

Using this method, gamma skin dose rates as high as 20 mrads per hour, or 200 times the authorized emission rates, were recorded. These values indicated some action had to be taken.

Regulatory Action.

A notice to manufacturers and distributors of radium dial pocket watches was issued by the Office of Radiation Control, which set forth the interpretations of the Mayor's Technical Advisory Committee on Radiation as previously stated. The notice requested that the offer for sale or supply of radium-dial pocket watches to retail stores within New York City be discontinued at once, and that

all existing stocks on consignment from retail outlets be withdrawn by February 1, 1963.

Following this, New York City Board of Health enacted the following amendment to the "Radiological Hazards" section of the Health Code, to become effective on February 1, 1963:

"A permit shall be required for any timepiece, instrument, novelty or device containing radioactive material other than hydrogen-3 (tritium), if the regular use thereof in the usual manner will expose the user to a radiation dose in excess of 0.5 rems per year to the head, trunk, active bloodforming organs, gonads, or lens of the eye, or will expose the hands or forearms to a dose in excess of 7.5 rems per year, when used regularly."

The new amendment does not necessarily mean that watches having excessive quantities of radium are absolutely prohibited from sale. Such a watch may be sold to anyone having a Health Department permit to possess one. This is to allow anyone who might have a legitimate professional need for such a watch to use one under the necessary safeguards prescribed in the terms of a license.

Summary

The New York City Department of Health has enacted an amendment to the Health Code, affecting timepieces containing radioactive material other than tritium, which considers the recommendations of the NCRP and the FRC. This has been interpreted by the Mayor's Technical Advisory Committee on Radiation to limit the radiation emission from the front of pocket watches to 0.1 mrad per hour, and from the back of wrist watches to 1.0 mrad per hour.

Rough field measurements of watches are made with an end-window G-M survey meter, but more precise measurements make use of autoradiographs on personnel dosimeter film.

Any person desiring the purchase a watch that

emits radiation in excess of the specified limits may do so by demonstrating a legitimate, professional need and securing a permit from the New York City Department of Health.

Acknowledgment

I wish to express my appreciation to Hanson Blatz, Director of the Office of Radiation Control. for making available his files and correspondence, which aided immeasurably in the preparation of this paper.

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Environmental Levels of Radioactivity of Atomic Energy Commission Installations

The U.S. Atomic Energy Commission receives from its contractors quarterly reports on the environmental levels of radioactivity in the vicinity of major Commission installations. The reports include data from routine monitoring programs where operations are of such a nature that plant perimeter surveys are required.

Summaries of the environmental radioactivity data for 22 AEC installations have appeared periodically in *Radiological Health Data* since November 1960. Summaries follow for Argonne National Laboratory, Portsmouth Area Gaseous Diffusion Plant, and Shippingport Atomic Power Station.

The measured concentration of a radionuclide in air and water may be compared with the Maximum Permissible Concentration (MPC) of that nuclide as recommended by the National Committee on Radiation Protection and Measurement (NCRP). For the environment near an AEC installation, the applicable MPC's are one-tenth of the occupational MPC values for continuous exposure given in the National Bureau of Standards "Handbook 69" (NCRP report no. 22). The MPC values applicable to the reports that follow are given in table 1.

In these reports, nonspecific terms such as "total activity," "total alpha," and "gross beta" do not in themselves suggest any one MPC value. Often, when concentrations are low a laboratory will assign an MPC value that is more restrictive than necessary. This avoids the more costly isotopic tests necessary to justify a less restrictive value. References to table 1 will be made to designate the appropriate MPC's adopted by the respective laboratories.

TABLE 1.—SELECTED ENVIRONMENTAL MPC VALUES PERTAINING TO AEC INSTALLATION REPORTS IN THIS SUBSECTION

		Environmental MPC		
Line No.	Radionuclide or mixture of radionuclides	Water (µµc/liter)	Air (µµe/m³)	
1	If Sr 80, I 129, Ph210, Po210, Ra223, Ra226, Ra226, Pa221, and Th-nat are not presents	2,000		
2	If Sr 90, Pb210, Ra226, Ra226 are not present*	600		
3	If Ra ²²⁶ , Ra ²²⁸ are not present ^a	100		
4	Mixture of unidentified nuclides	10	0.0	
5	If α emitters and Ac ²²⁷ are not present*		1.	
6	If α emitters and Pb216, Ac227, Ra228, and Pu241 are not presents.	-	10	
7	If α emitters and Sr 90, I 129, Pb210, Ac227, Ra229, Pu241, and Bk240 are not present*		100	
8	Barium-lanthanum-140	20,000	4,000	
9	Cerium-141	90,000	5,000	
10	Cerium-144	10,000	300	
11	Cesium-137	20,000	500	
12	Cobalt-58	90,000	2,000	
13	Cobalt-60 Hydrogen-3 (tritium)	50,000	300	
14	- T.	3,000,000	500,000 300	
15 16	lodine-131 Plutonjum-239	2,000 5,000	0.	
17	Ruthenium-103	80,000	20.000	
18	Ruthenium-rhodium-106	10.000	20,000	
19	Strontium-80	10,000	1.000	
20	Strontium-90	100	1,000	
21	Thorium-232	2,000	10	
22	Thorium-protactinium-234	20,000	2.000	
23	Uranium, natural	20,000	2,000	
24	Xenon-133	20,000	300	
25	Zirconjum-njobjum-95	60.000	1.000	

[&]quot;Not present" implies the concentration of the nuclide is small compared with its appropriate MPC. According to AEC regulation (Federal Register, Title 10, Part 20, August 9, 1961), a group of nuclides may be considered not present if the ratio of each nuclide to its appropriate MPC is equal to or ess than 1/10 and if the sum of these ratios for the group in question is equal to or less than 1/4.

ARGONNE NATIONAL LABORATORY 1962

University of Chicago, Lemont, Illinois Air Monitoring

Weekly continuous air filter samples were collected at seven locations on the Argonne National Laboratory (ANL) site as shown in figure 1 and at 5 off-site locations at Aurora (west of ANL site), Wheaton (northwest), Hinsdale (northeast), Joliet (southwest), and Tinley Park (southwest). The quarterly averages of alpha, beta, and several nuclide concentrations are given in table 2. The radionuclide determinations were made by gamma spectra of manthly composites of the filters. The data show little difference between off-site and onsite measurements for alpha activity and most of the nuclides, inferring that ANL does not contribute detectable quantities of these activities to the atmosphere.

Water Monitoring

ANL waste water is discharged into Sawmill Creek, a stream that runs through the Argonne grounds and enters the Des Plaines River about 500 yards downstream from the waste water discharge. Sampling locations on Sawmill Creek

and Des Plaines River are shown in figures 1 and 2 respectively.

On Sawmill Creek, weekly grab samples are collected upstream and three times a week samples are collected downstream from the waste water outfall. The upstream flow is roughly equal to the waste water flow, yielding a dilution factor of one-half. The data in table 3 show significantly higher concentrations downstream than upstream, an indication of the radioactivity contributed to the stream by ANL. A comparison of the downstream concentrations with the environmental MPC's listed in table 1 shows that all concentrations are relatively low.

Weekly grab samples are collected from the Des Plaines River upstream and downstream from its junction with Sawmill Creek. The results indicate that the dilution factor of the Des Plaines River is so large that the radioactivity contribution from ANL was not detected.

Previous coverage in Radiological Health Data:

Period	Issue
1959 and first quarter 1960	December 1960
Second quarter 1960	April 1961
Third and fourth quarters 1960	July 1961
First and second quarters 1961	December 1961
Third and fourth quarters 1961	May 1962

TABLE 2.—RADIOACTIVITY OF AIRBORNE PARTICULATES, ANL 1962

[Concentrations in µµc/m³]

	Sampling	ampling				
Type of analysis	locations	1st quarter	2nd quarter	3rd quarter	4th quarter	average
Alpha	on-site off-site	0.0036 0.0039	0.0054 0.0052	0.0049 0.0051	0.0055 0.0060	0.004
Beta	on-site off-site	5.1 4.9	5.5 5.1	3.4	5.5 5.7	4.9
Ba-La ¹⁴⁰	on-site off-site	a < 0.01 a < 0.01	0.032 0.025	0.60 0.66	1.5	0.56 0.61
Ce 141	on-site off-site	0.27 0.23	0.36 0.39	0.31 0.29	0.46	0.36 0.34
Ce ¹⁴⁴	on-site off-site	0.85 1.0	1.6 1.5	1.0 1.0	1.5	1.3
Cs137	on-site off-site	0.023 0.034	0.090 0.065	0.040 0.042	0.028 0.030	0.046 0.043
Ru ¹⁰³	on-site off-site	0.38 0.43	0.23 0.24	0.21 0.26	0.44	0.32 0.36
Ru-Rh ¹⁰⁶	on-site off-site	0.29 0.30	0.50 0.51	0.39 0.37	0.30	$0.38 \\ 0.38$
Zr-Nb ⁰⁸	on-site off-site	1.9	1.6	0.86	1.3	1.4

a Not analyzed during January and February.



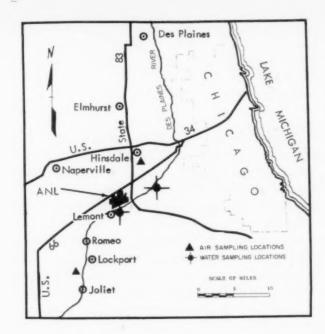


FIGURE 1.—ON-SITE SAMPLING LOCATIONS, ARGONNE NATIONAL LABORATORY

 $\begin{array}{ccccc} {\bf Figure} & 2. \\ \hline {\bf CSITE} & {\bf LOCATION} & {\bf OF} & {\bf ARGONNE} & {\bf NATIONAL} \\ & {\bf LABORATORY} & ({\bf INCLUDING} & {\bf SOME} & {\bf OFF-} \\ & & {\bf SITE} & {\bf SAMPLING} & {\bf STATIONS}) \end{array}$

TABLE 3.—RADIOACTIVITY IN SAWMILL CREEK, ANL, 1962

		Concentrations (µµc/liter)				
Type of analysis	Sampling locations	Maximum	Minimum	1962 Average		
Alpha emitters:	upstream	4.1	0.2	1.6		
Total alpha	downstream	16.4		5.0		
U-natural	upstream downstream	1.9 7.6	0.4	$\frac{1.0}{2.8}$		
Pu ²³⁹	upstream	<0.05	<0.05	<0.05		
	downstream	0.91	<0.05	0.12		
Th ²³²	upstream	0.10	<0.05	<0.05		
	downstream	0.48	<0.05	0.12		
Beta emitters:	upstream	393	10	40		
Total beta	downstream	806	17	66		
Co ⁵⁸	upstream	<5	<5	< 5		
	downstream	3,260	<5	136		
Co ⁶⁰	upstream	<3	<3	<3		
	downstream	75	<3	6		
Sr ⁸⁹	upstream	56	<2	12		
	downstream	25	<2	9.5		
Sr ⁹⁰	upstream downstream	2.4 3.1	<0.5 <0.5	$\frac{1.2}{1.2}$		
I ¹⁵¹	upstream	<3	<3	<3		
	downstream	11	<3	<3		
Cs ¹³⁷	upstream	7.7	<0.5	1.8		
	downstream	254	<0.5	7.6		
Ba ¹⁴⁰	upstream downstream	3.4	<1 <1	<1 <1		
Th-Pa ²³⁴	upstream downstream	1.5	0.3	0.8		

PORTSMOUTH AREA GASEOUS DIFFUSION PLANT 1962

Goodyear Atomic Corporation Portsmouth, Ohio

The separation of uranium isotopes by the gaseous diffusion process presents control problems similar to any chemical process using toxic solvents and extraction solutions. Natural uranium and thorium-234 are the most likely radionuclides to be released to the environment by the Portsmouth Area Gaseous Diffusion Plant. Since natural uranium is an alpha emitter and thorium-234 is a beta-gamma emitter, environmental monitoring is conducted for evidence of alpha and beta-gamma emitters to test the effectiveness of plant controls.

Air samples are collected monthly at 17 sites located from 1 to 6 miles from the plant as shown in figure 3. Monthly water samples are collected at 14 locations within 5 miles of the plant.

Average alpha and beta-gamma concentrations in air and water are summarized in table 4. The external gamma levels are measured at the air sampling locations shown in figure 3 and the results included in table 4.

Water alpha and beta-gamma concentrations remain essentially unchanged from previous quarters. The slight increase in external gamma dose rate is evidently a consequence of nuclear weapons testing fallout. Total amount of radiation detected was too small to determine the amount, if any, attributable to plant operations.

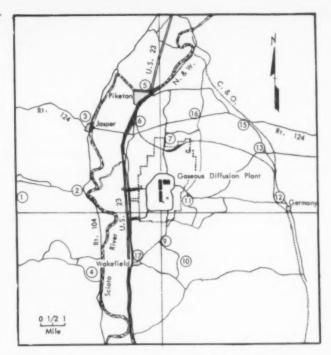


FIGURE 3.—AIR SAMPLING LOCATIONS, PORTS-MOUTH GASEOUS DIFFUSION PLANT

Previous coverage in Radiological Health Data:

Period	Issue
1959 and first quarter 1960	November 1960
Second and third quarters 1960	March 1961
Fourth quarter 1960	August 1961
First and second quarters 1961	February 1962
Third and fourth quarters 1961	September 1962

TABLE 4.—ENVIRONMENTAL RADIOACTIVITY, PORTSMOUTH PLANT, 1962

Basis of measurement	Number of samples	Unit	Maximum	Minimum	1962 average
Air (alpha concentration) Air (beta-gamma concentration) Water (alpha concentration) Water (beta-gamma concentration) External gamma	180 180 162 162 180	μμc/m³ μμc/m³ μμc/liter μμc/liter mrad/hr	0.7 39.3 1,105 1,030 0.076	0.1 0.1 0.5 14 0.015	0.1 5.7 25.6 61 0.038

SHIPPINGPORT ATOMIC POWER STATION Second Half 1961 and First Half 1962

Duquesne Light Company, Shippingport, Pennsylvania

Environmental radiation monitoring at the Shippingport Atomic Power Station began with a two-year preoperational survey program to establish background levels at the site of the world's first large-scale nuclear-powered electric generating station. Following initial operation of the plant in December 1957, this program was continued as orginally conceived through the third quarter of 1961, when it was determined that fewer sampling locations closer to the plant would provide equal or better evaluation of the effects of plant operation on the environment.

Area Monitoring

Reduced air monitoring is apparent from an examination of figure 4. Only two on-site monitoring trailers currently operate at the eastern and western boundaries, and one off-site unit operates at Midland. Air samples are collected continuously at each trailer and monitored for gross beta activity. Average concentrations of radioactivity in air for the last half of 1961 and the first half of 1962 are presented in table 5.

Beta-gamma radiation levels are also continuously monitored at these stations. There was no significant difference in average levels measured during the year under consideration from those of previous years. The average beta-gamma level for the last half of 1961 was 0.015 mrem/hr, and during the first half of 1962 the average beta-gamma level was 0.022 mrem/hr.

Until December 1961, fallout samples were collected weekly on gummed paper trays. No fallout samples were collected during December, but collections have been made monthly at the three monitoring stations since collection pots were placed in service on December 29, 1961. Fallout gross beta activity data for the year ending June 30, 1962 are summarized in table 6. Increased fallout levels during this time are attributed to fallout from weapons tests since there was no indication of any plant contribution.

Release of Radioactive Wastes to the Atmosphere

An incinerator for burning contaminated combustible material is located in the waste disposal

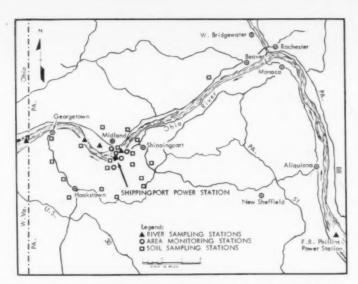


FIGURE 4.—SHIPPINGPORT POWER STATION SAMPLING LOCATIONS

TABLE 5.—AIRBORNE PARTICULATE RADIOACTIVITY, SHIPPINGPORT

[Average concentrations in \(\mu\mu\c/m^3\)]

Values	Third quarter 1961	Fourth quarter 1961	Calendar year 1961	First half 1962
Minimum	0.08 8.0	0.08 7.1	0.06 7.5	0.08
Average	1.5	1.5	1.2	0.89

plant. The exhaust from the incinerator passes through a wet gas scrubber and a filter before entering the stack.

During the third quarter of 1961, a total of 4,636 microcuries of gaseous radioactive waste (primarily xenon-133) was released to the atmosphere at a controlled rate over a period of 93 hours and 27 minutes. During the fourth quarter a total of 13,993 microcuries of gaseous radioactive waste (primarily xenon-133) was released to the atmosphere at a controlled rate over a period of 183 hours, the average concentration at the stack exit during release being 5,000 $\mu\mu c/m^3$. For 1961 as a whole, the release of these wastes totaled 103.5 millicuries, and the concentration at the stack exit during release average 8,100 μμc/m³. There was no release of gaseous radioactivity to the atmosphere from the waste disposal system during the first half of 1962.

Liquid Radioactive Waste Monitoring

Tritium (H³) is released periodically in controlled quantities and concentrations to the Ohio River. Toward the end of 1960, the ion exchange resin in

TABLE 6.—GROSS BETA ACTIVITY IN FALLOUT, SHIPPINGPORT

[Average concentrations in mc/km3/month]

Sampling locations	Third quarter 1961	Fourth quarter 1961	Calendar year 1961	First half 1962
Upwind: ½ mile SW of site ½ mile NW of site. Downwind: On site SE of main bldg ½ mile NE of site. Average of all stations (except SW upwind)	3.8 12.1 9.0 199	^b RFS 187 6.3 149	1.9 50.5 525 39.4 47.4	e e e e
Pot No. 1 ^d Pot No. 2 Pot No. 3	=	=	=	52 . 39 . 50 .

[•] One mc/km² = 2.59 mc/mi². b Removed from service. c Discontinued in 1962. d Collection of fallout in pots began December 29, 1961.

the reactor coolant purification system was changed from a natural lithium hydroxyl form to a lithium-7 enriched form. This change resulted in a decrease in tritium production within the reactor by a factor of about 30. The average total daily tritium activities released to the Ohio River during the last two quarters of 1961 were 7,000 and 4,000 respectively. The average tritium concentrations in the plant effluent channel during these periods were 53 $\mu\mu$ c/liter and 37 $\mu\mu$ c/liter, respectively. During the first half of 1962, the average total daily tritium activity released to the Ohio River was 3,900 μ c and the average concentration in the plant effluent was 39 $\mu\mu\mu$ c/liter.

Gross radioactivity of unidentified nuclides (does not include tritium) discharged during the last half of 1961 averaged 493 µc/day in the third quarter and 415 µc/day in the fourth quarter. Average concentrations in the effluent channel during release in these periods were 4.0 and 3.2 μμc/liter, respectively. During the first half of 1962, gross radioactivity of unidentified nuclides averaged 222 µc/day and totaled 40,323 microcuries. Monthly average concentrations in the effluent channel during releases in this period ranged from 1.5 to 4.2 µc/liter. These concentrations are in addition to normal background radioactivity in the condenser cooling water used for dilution in the effluent channel prior to discharge to the Ohio River.

The revised liquid radioactive waste sampling program became effective in the fourth quarter of 1961 and the first quarter of 1962 as follows: River water sampling was reduced by the end of the first quarter 1962 to weekly samples taken continuously at the plant intake and outfall to the

Ohio River. These samples are analyzed for gross alpha and gross beta activity in suspended and dissolved solids and for total potassium-40 activity. After the third quarter of 1961, sampling at Phillips power Station and Dam No. 7 was discontinued, and by the end of the first quarter of 1962 sampling was also discontinued at the midland and East Liverpool stations (see figure 4); however the Pennsylvania State Health Department assumed responsibility for sampling and analytical programs at the latter two stations. Results for the last half of 1961 and the first half of 1962 are summarized in table 7.

Other Sampling Activities

Several programs of environmental sampling are of some interest to the Pennsylvania Department of Health. Two of these are currently active; a third, that of sampling vegetation and aquatic life, has been eliminated. River sediment and silt sample collections are made semiannually and sent as directed to the State Health Department for analysis. A series of ten soil samples are collected each year to be retained for later analysis if such data is required.

Previous coverage in Radiological Health Data:

Period	Issue
1959	July 1960
First quarter 1960	December 1960
Second quarter 1960	January 1961
Third and fourth quarters 1960	October 1961
First and second quarters 1960	April 1962

TABLE 7.—GROSS ALPHA AND BETA CONCENTRATIONS IN THE OHIO RIVER

[Average concentrations in \(\mu\mu\epsilon\)/[iter]

Sampling station	Third quarter 1961		Fourth quarter 1961		Calendar year 1961		First half 1962	
	Alpha	Beta	Alpha	Beta	Alpha	Beta	Alpha	Beta
Jpstream: Phillips Power Station. Shippingport intake	1.0	12.4 17.7	4.9	31.8			3.9	38.8
Downstream: Shippingport outfall	1.2 1.8 1.4	27.7 15.8 12.5 11.5	1.2 3.2	25.9 34.8 24.3	b a	b a b	2.0	42.

Station discontinued sampling after third quarter 1961.
 Monitored separately by the Pennsylvania Department of Health since the end of 1961.

Reported Nuclear Detonations

APRIL 1963

Only one nuclear detonation was announced by the Atomic Energy Commission during the month of April 1963. Abritrarily referenced by Radiological Health Data as test number 102, this test of low yield range was conducted underground at the Nevada Test Site on April 5. (Low yield range has been announced as being less than 20 kilotons yield.)

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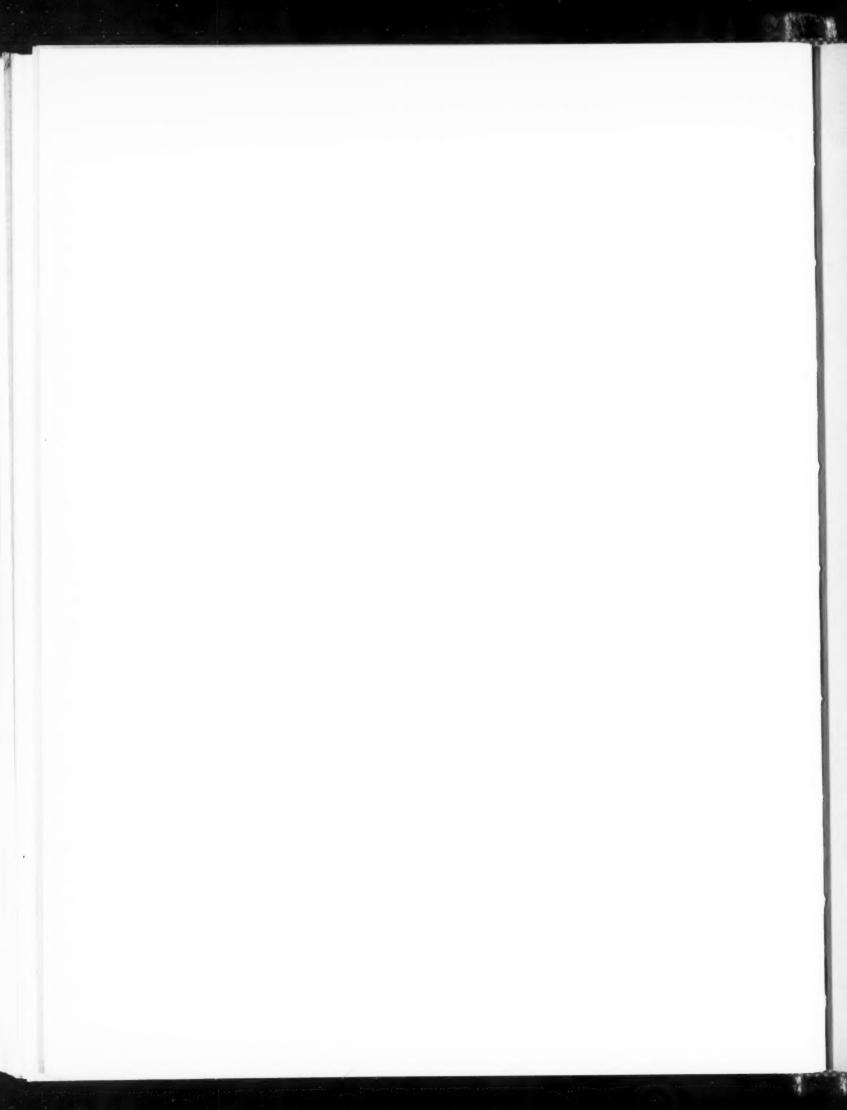
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UNITS AND EQUVALENTS

For the convenience of the Radiological Health Data (RHD) reader a selected list of units and equivalents is presented below.

Symbol	Name	Equivalent
Bev	billion electron volts	
pm	count per minute	
lpm	disintegration per minute	
	gram	
(g	kilogram	1 kg = 1000 gm = 2.2 pounds
cm ²	square kilometer	
cvp	kilovolt peak	
n³	cubic meter	1 m ³ = 1000 liters
na	milliampere	
nas	milliampere-second	
Mev	million electron volts	
mi ²	square mile	
ml	milliliter	
nm	millimeter	precipitation:
		$mm = \frac{m\mu c/m^2}{\mu\mu c/liter} \times 1000 = \frac{liter}{m^2}$
mrad	millirad	
mrem	millirem	
mr/hr	milliroentgen per hour	
тµс	millimicrocurie	$1 \text{ m}_{\mu}\text{c} = 1 \text{ nc}$
nc	nanocurie	$1 \text{ nc} = 1000 \text{ pc} = 1 \text{ m}_{\mu}\text{c} = 10^{-6} \text{ curies}$
nc/m ²	nanocurie per square meter	$1 \text{ nc/m}^2 = 1 \text{ m}_{\mu}\text{c/m}^2$
		= 1,000 $\mu\mu c/m^2$ = 1 mc/km ² = 2.59 me/mi ²
pc	picocurie	$1 \text{ pc} = 1 \mu \mu c = 10^{-12} \text{ curies}$
r	roentgen	
	micromicrocurie	$1 \mu \mu c = 2.22 \text{ dpm}$

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